

Structure, Morphology and Luminescent Properties of

YVO₄: Dy³⁺, Bi³⁺ Phosphors

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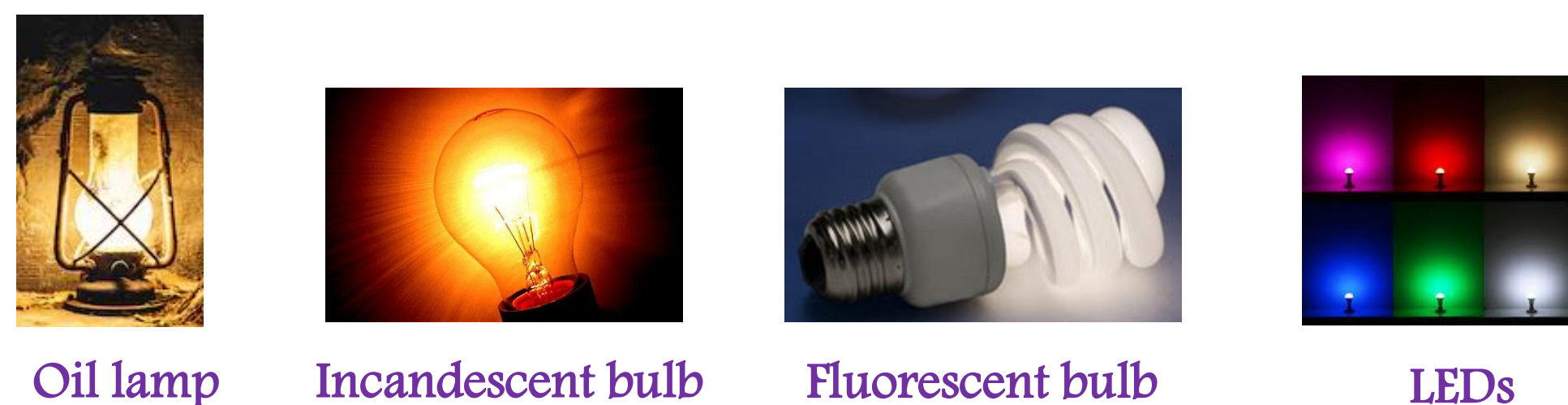
ABSTRACT

Dy³⁺- Bi³⁺ co-doped YVO₄ (YVO₄: 3at%Dy³⁺, xBi³⁺) phosphors have been synthesized and subsequently characterized by XRD, SEM, photoluminescence and Raman spectroscopy to study the simultaneous effect of the activator and sensitizer on the luminescent properties of said phosphor.

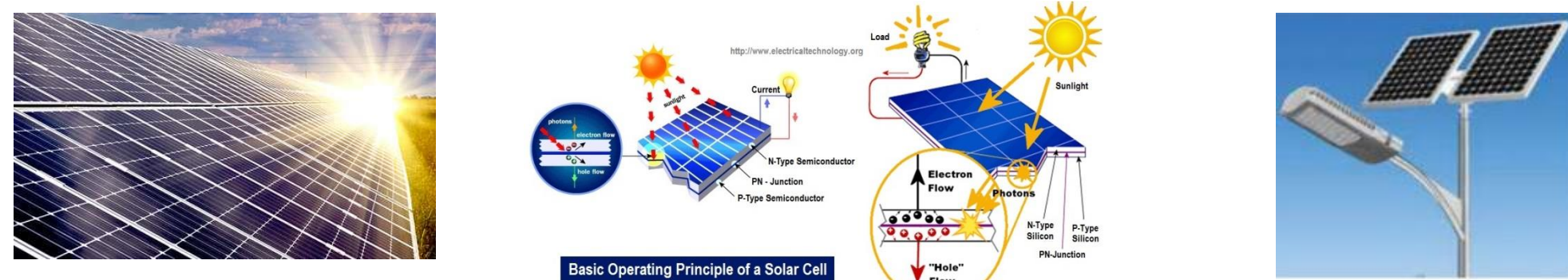
The observations in this case were subsequently explained on the basis of change in the local symmetry of the dopant ion, due to change in the host morphology.

INTRODUCTION

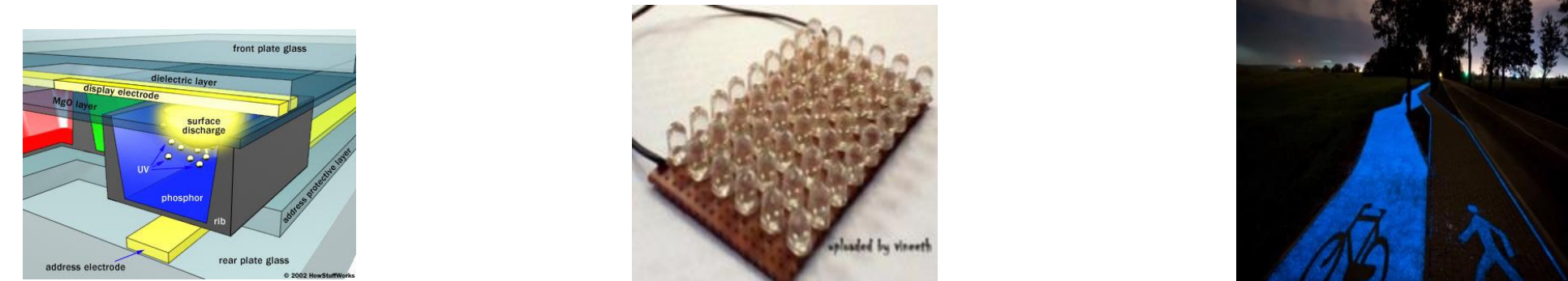
History of Light



- ❖ Luminescence is a cold light.
- ❖ Materials exhibiting this phenomenon are called Phosphors.
- ❖ The research on rare earth (RE) doped luminescence materials is always the most active field owing to the large need in the field of illuminations.



Solar Cells



Plasma Displays

LED Solar cells

- ❖ Their wide application include solar Cells, plasma displays, LEDs, CRTs, fluorescent lamps, etc.
- ❖ YVO₄ is a promising host material for various optical applications, especially doped with trivalent rare earth cations.
- ❖ Dy doped YVO₄ is considered as potential phosphors due to its high absorption and luminescence efficiencies in UV region.

EXPERIMENTAL

YVO₄:Dy³⁺,xBi³⁺ powder samples were prepared by solid state route.

Y₂O₃, V₂O₅, Dy₂O₃, Bi₂O₃

ground for 6hrs

calcined at 900°C, 6hrs

sintered at 1100°C, 6hrs

YVO₄: 3%Dy³⁺, xBi³⁺ (x = 0,3,5,7)

➤ XRD: Rigaku, Japan (with Cu-Kα)

➤ SEM: JEOL, USA

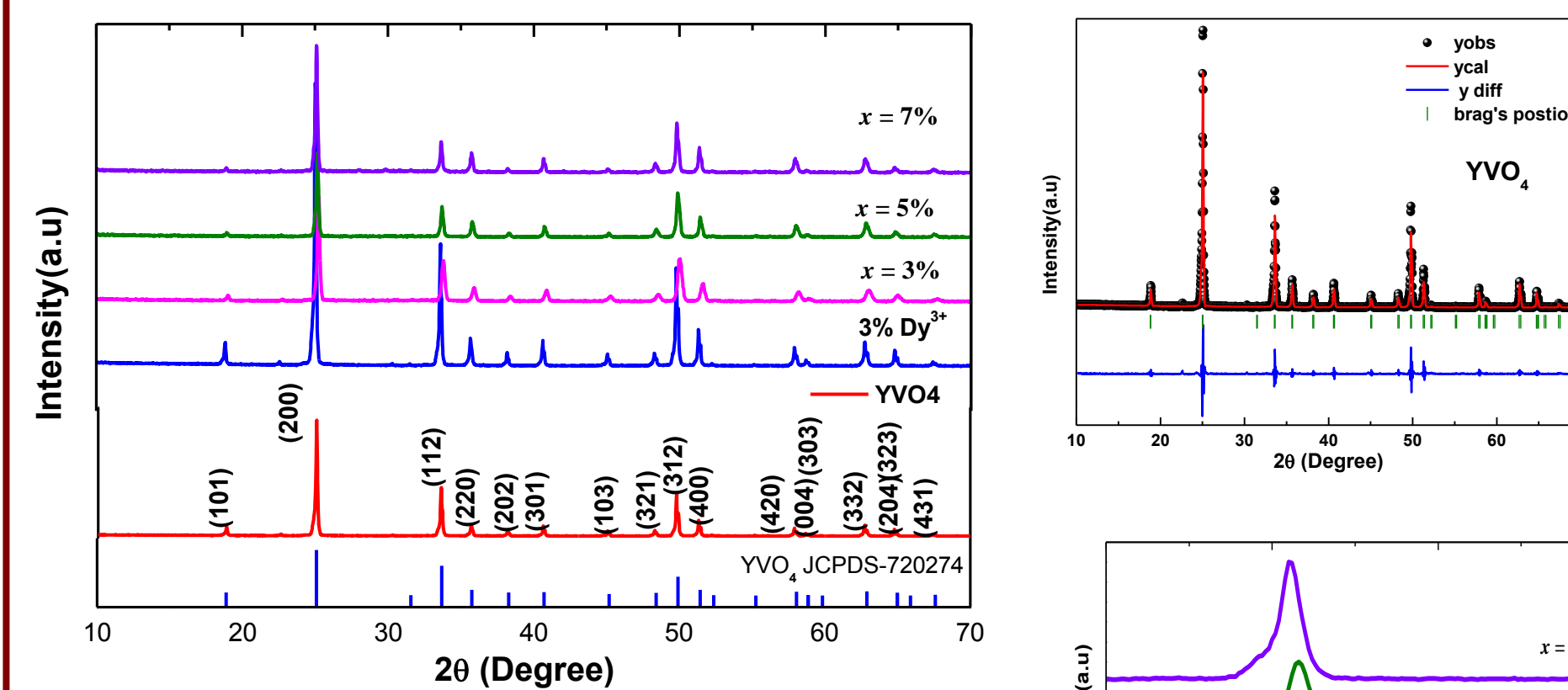
➤ RAMAN: Witech Micro Raman Spectrometer

➤ PL: Perkin Elmer LS55

RESULTS AND DISCUSSION

XRD

(YVO₄: 3%Dy³⁺, xBi³⁺)



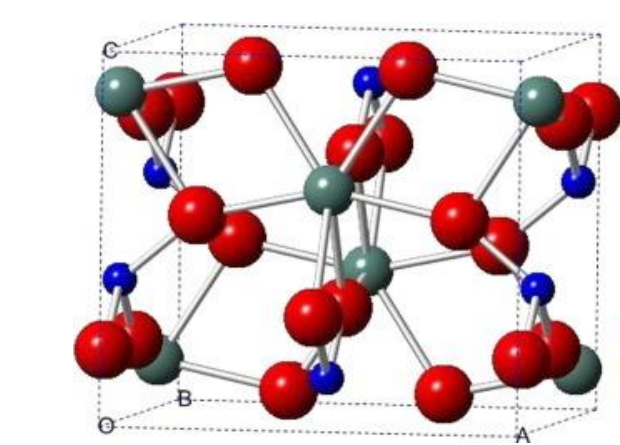
➤ Figure shows the few representative XRD patterns of the YVO₄: Dy³⁺, Bi³⁺ phosphors annealed at 1100°C.

➤ The exact structure of the samples, Rietveld analysis of the XRD data of the parent samples is carried out by using Fullprof program.

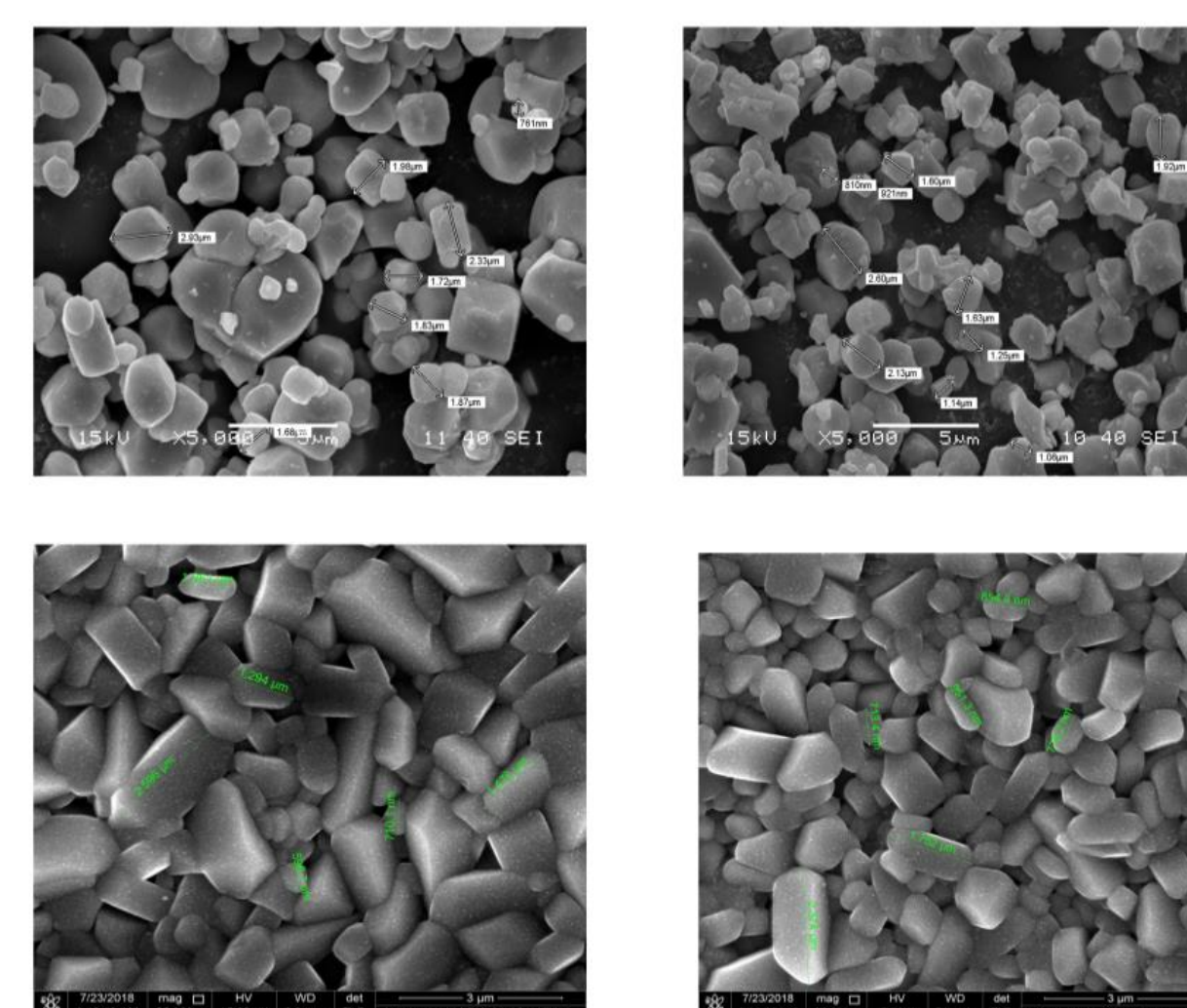
➤ Samples exhibited single phase tetragonal structures in space group *I141/amd* which is also consistent with the JCPDS Card No. 72-0274 of pure crystalline tetragonal YVO₄.

➤ With increase in the Bi³⁺ concentration in YVO₄ samples, the (200) diffraction peak shifts to lower angle side, might be due to the reason that the ionic radius of Bi³⁺ ion is slightly higher than that of Y³⁺ ion.

➤ The crystal structure shown here is obtained from the data obtained from the refinement of XRD data, where the subsequent samples are co-doped with Dy and Bi.



SEM



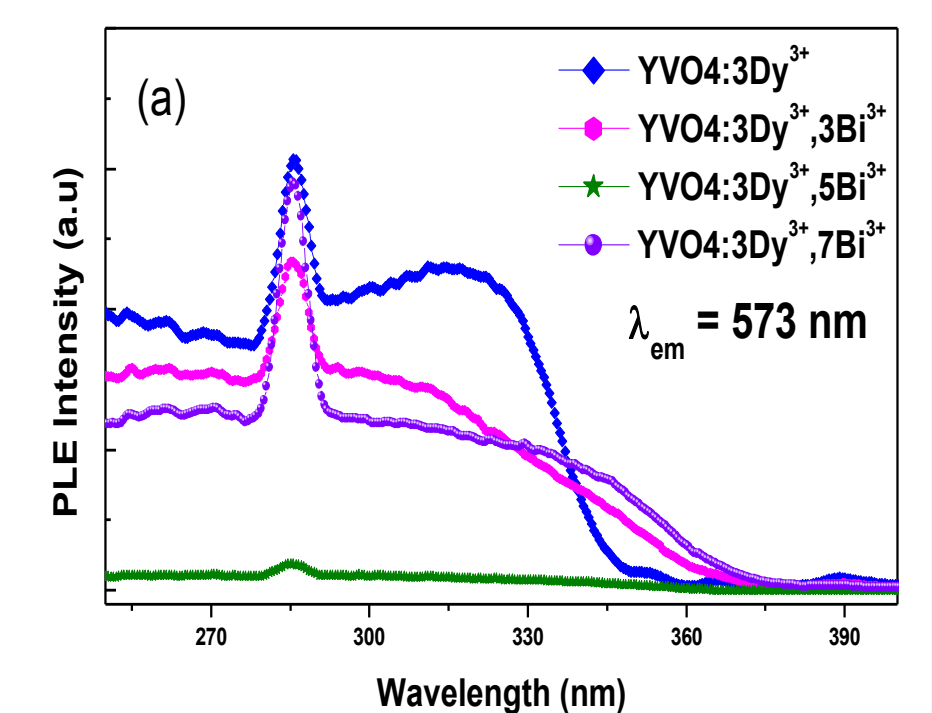
➤ SEM investigations were carried out to understand surface morphological features and the particle size of the phosphor.

➤ The SEM micrograph clearly exhibits homogeneous and smooth surfaced grains having almost hexagonal structure.

➤ The average particle size after heat treatment is found to be ~ 1 μm and this microcrystalline form of crystalline phosphors may result in high luminescent intensities.

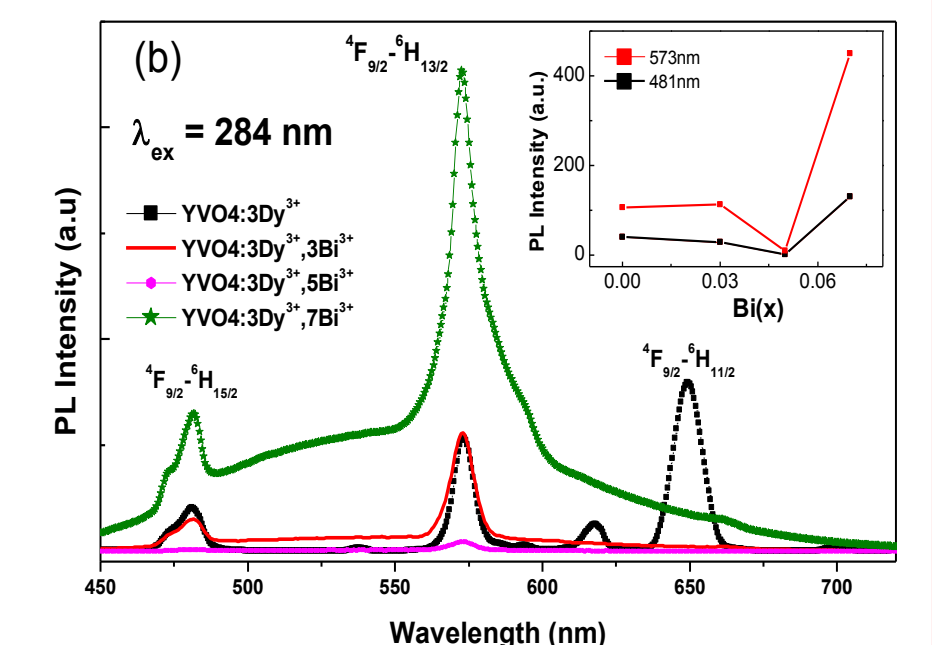
PHOTOLUMINESCENCE

Figure shows the PLE and PL spectra of the Dy singly doped YVO₄ and co-doping of Dy³⁺ and Bi³⁺ (inset shows the I_{max} of emission peaks with Bi³⁺ conc.).



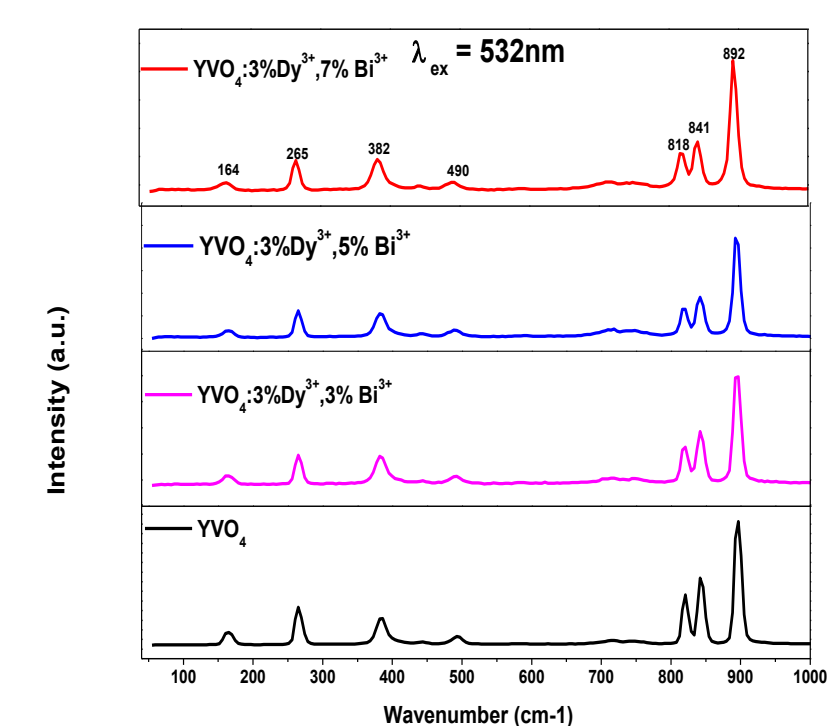
573 nm peak, initially increases with Bi and reaches a maximum for 7at%.

May be due to conc. quenching of RE ions at lower %, which is further compensated by higher Bi.



RAMAN SPECTROSCOPY

Figure shows the Raman spectra for Bi³⁺ (0, 3, 5, 7 at%) co-doped YVO₄ under 532 nm laser excitation.



Strong Raman peaks imply stronger interaction between the atoms, mainly arising from stretching and bending of the shorter metal-oxygen bond within anionic groups.

CONCLUSION

One significant outcome of the photoluminescence study involved a drastic variation of relative intensity due to 7at% of Bi³⁺ and the variation in the relative intensity of the ⁴F_{9/2} - ⁶H_{13/2} transition with respect to the ⁴F_{9/2} - ⁶H_{15/2} transition of Dy³⁺ in YVO₄: Dy³⁺.

ACKNOWLEDGEMENT

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