

Generation of Dimethyl Sulfoxide Coordinated Thermally Stable Halogen Cation Pools for C-H Halogenation



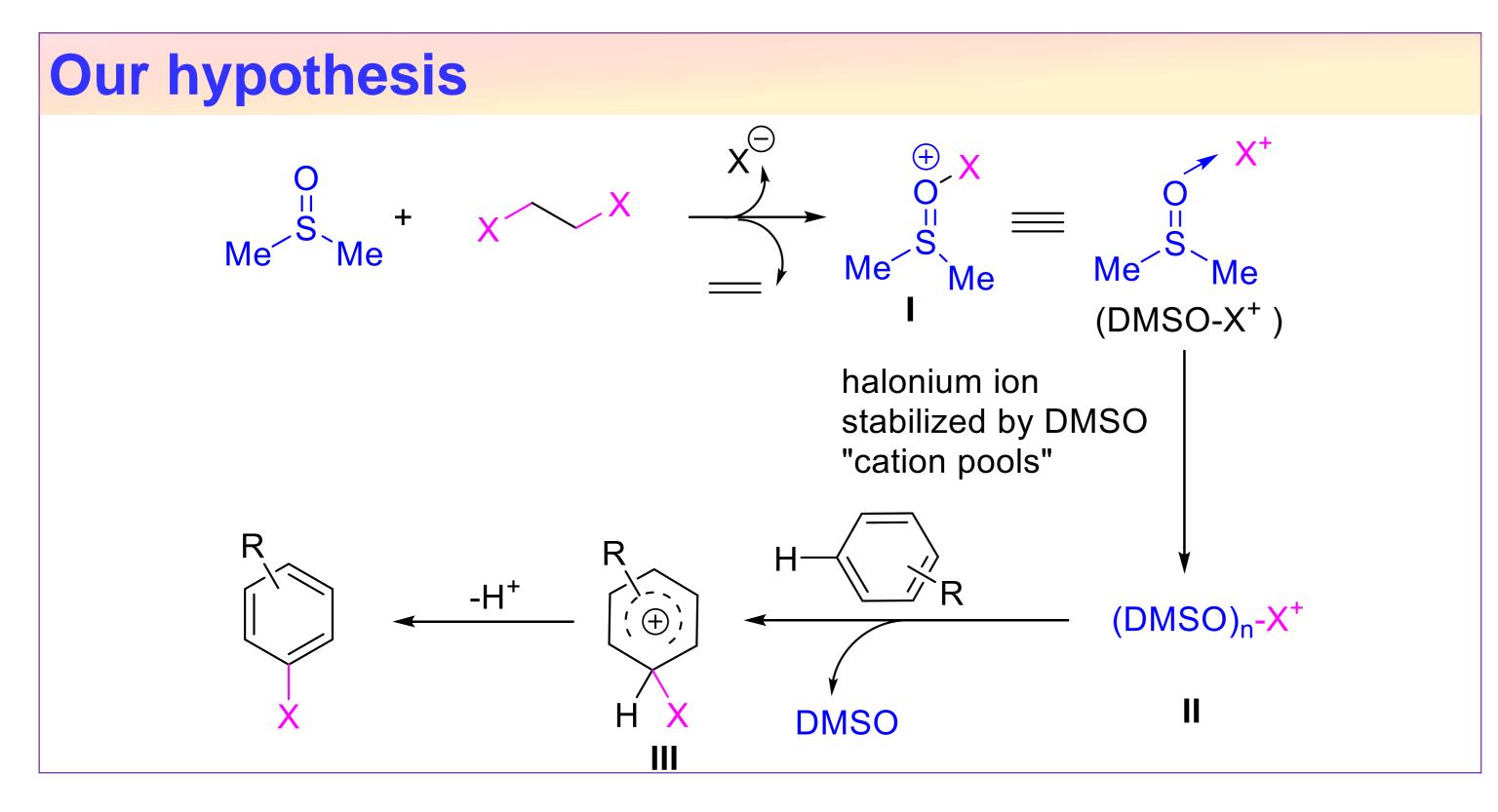
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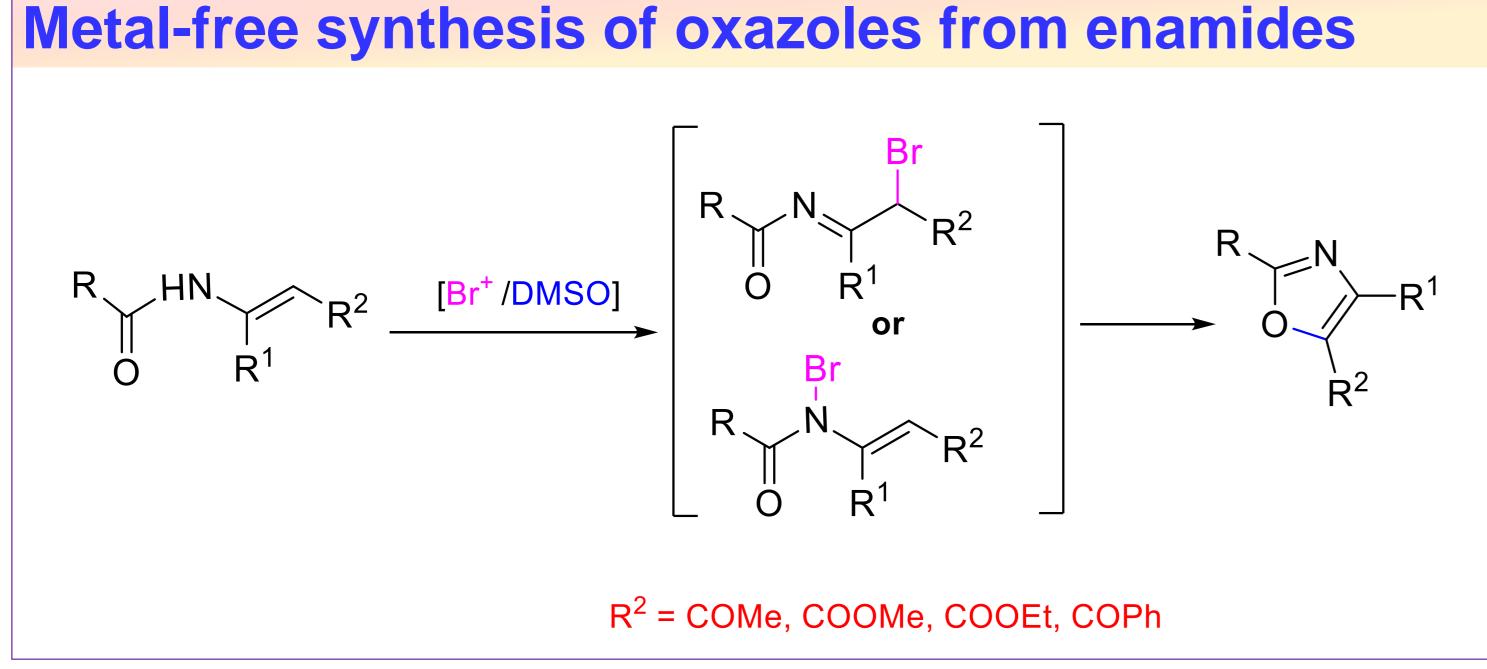
Abstract:

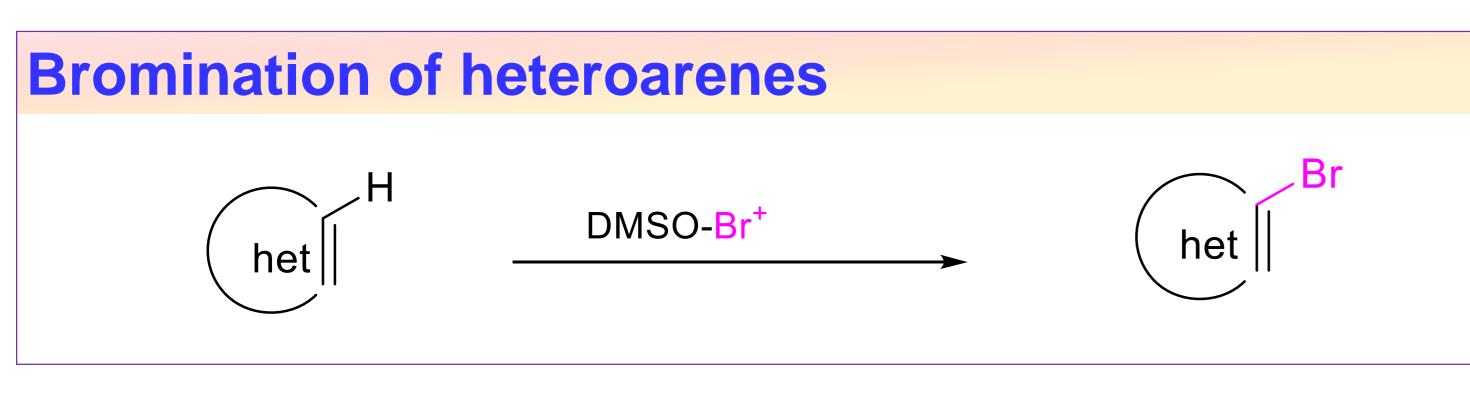
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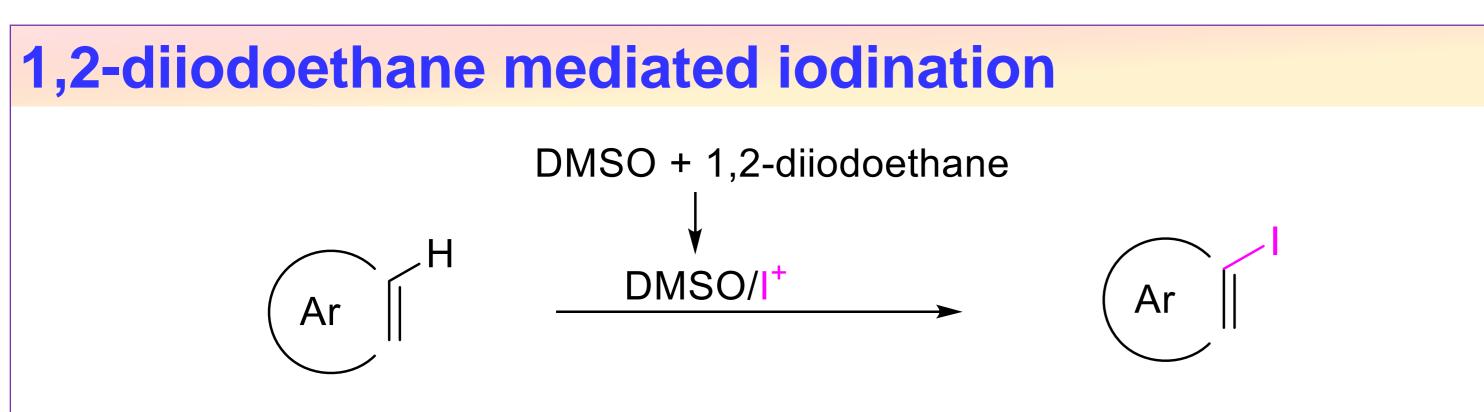
A method to generate halogen cation pools from the reaction of 1,2-dihaloethanes (hal= Br, I) and dimethyl sulfoxide (DMSO) for C-H halogenation of arenes and heteroarenes was reported. The initial reaction of DMSO and 1,2-dihaloethane generates the sulfur ylide, which undergoes pyrolytic elimination of ethylene by affording halonium ions. These ions were accumulated and stabilized by DMSO through coordination by forming halogen cation pools for the halogenation reaction. This protocol was selective for electrophilic monohalogenation of arenes at room temperature; however, polyhalogenated products were formed by raising the reaction temperature. Late-stage halogenation of heteroarenes and some commonly marketed drugs signifies the synthetic utility of this protocol in pharmaceutical chemistry. Unlike the classical methods, the in-situ generated electrophilic bromonium ion was further exploited for the direct synthesis of α -diketones from the alkenes under base-free conditions.

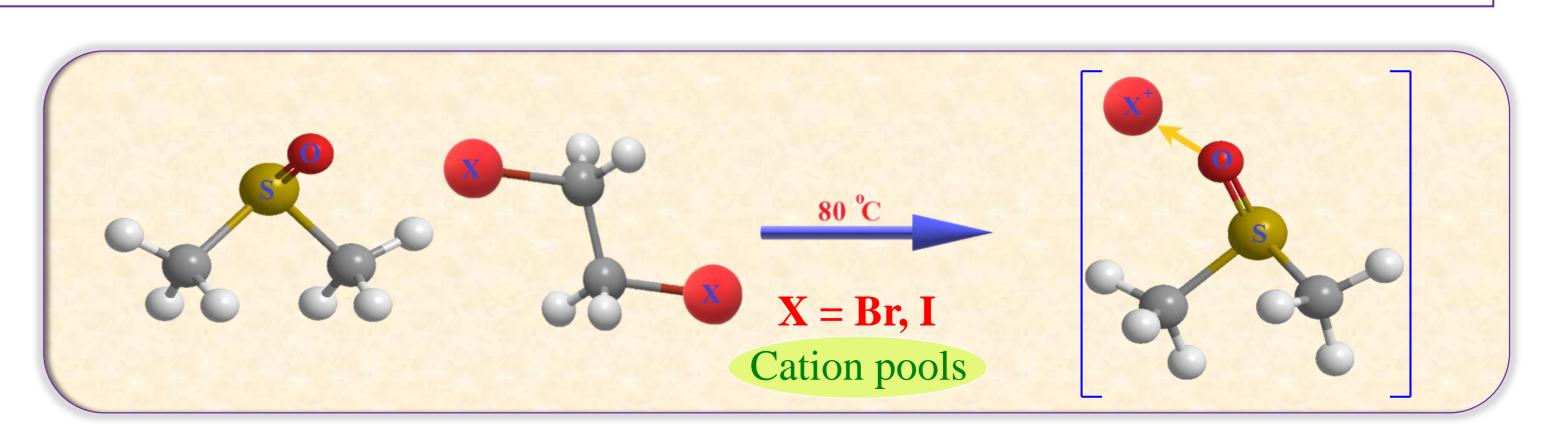


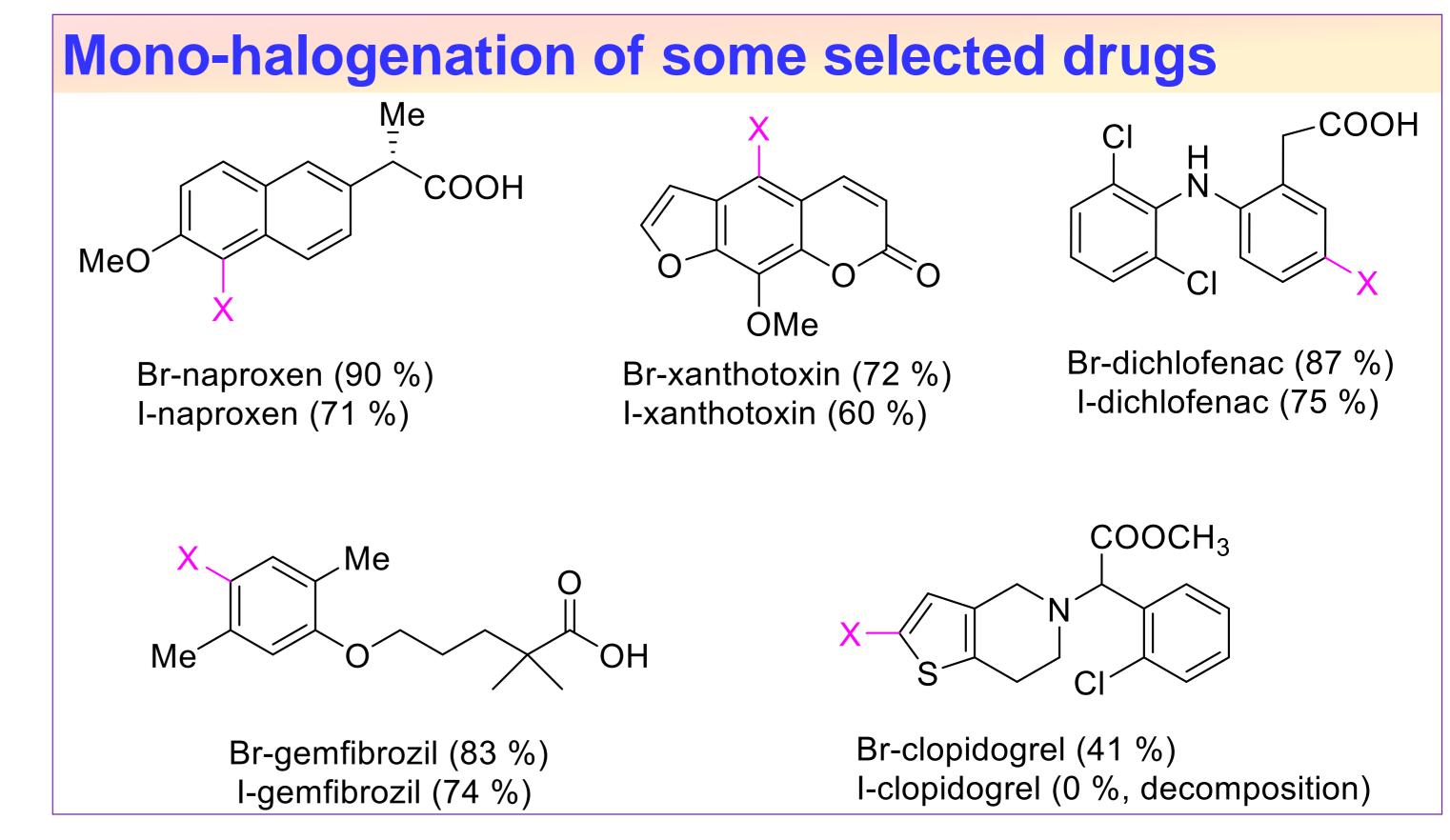
DBE/DMSO mediated C-H bromination DBE (2.5 mmol) 80 °C DMSO-Br⁺ DMSO (0.5 mL) Bromine "cation pools" The control of th











Reaction of olefin with bromine cation pools Product Pr

Conclusions

- □ Exploited the combination of dihaloethane (hal= Br, I)and DMSO as a precursor for generating thermally stable halogen cation pools (DMSO-X+).
- ☐ Using this preformed bromonium ion enamides proceeds intramolecular annulation to afford oxazoles.
- \Box This protocol is also efficient to produce dibromoalkane from alkylolefin and α-diketone from diaryl alkenes.
- ☐ The mild reaction conditions and operational simplicity allow the late-stage bromination and iodination of biologically potent heterocycles and drug candidates.
- ☐ 132 Substrates.

References

P.G. Dalai, K. Palit, N. Panda, *Adv. Synth. Catal.* **2022**, *364*, 1031.

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