

# Synthesis and characterization of pulsed electrodeposited Cu-Y<sub>2</sub>O<sub>3</sub> coating

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In the present study electrodeposited Cu-Y<sub>2</sub>O<sub>3</sub> composite coatings on copper substrate were synthesized from acidic copper sulfate bath with different concentrations (10 and 30g/l) of Y<sub>2</sub>O<sub>3</sub> ultrafine particles using direct and pulsed direct current mode for better surface-mechanical and oxidation properties. The study was aimed for improvement of surface-mechanical and oxidation behavior of the intended coatings compared to pure copper with retention of considerable electrical conductivity for possible electrical applications.

Phase/ morphology and elemental analysis of the coatings were carried out by XRD, SEM and EDS analysis. Surface-mechanical properties of deposited samples were studied systematically by microhardness and ball-on-plate type wear test. Thermal oxidation of the coating was also carried out. The coated samples' electrical conductivity was analyzed by four probe technique.

From the results, it was revealed that higher Y<sub>2</sub>O<sub>3</sub> content in electrolyte tends to increase Y<sub>2</sub>O<sub>3</sub> content in the deposits, which results in better hardness and wear resistance. Higher hardness and better wear resistance properties of co-deposited coatings were also attributed to finer matrix resulted by pulsing mode with higher pulse frequency. Better oxidation resistance was recorded in composite coatings due to better microstructure. Minor decrease in electrical conductivity of composite coatings was observed due to presence of Y<sub>2</sub>O<sub>3</sub> and finer matrix, but values were in acceptable range. The present method can be considered useful to improve surface properties of electrically used copper components.

**Key words:** Electro-codeposition; Cu; wear; oxidation; electrical conductivity.

## References:

- L. Tian, J. Xu, Electrodeposition and characterization of Ni-Y<sub>2</sub>O<sub>3</sub> composite, Appl. Surf. Sci. 257 (2011) 7615–7620.
- H.S. Moharana, A. Basu, Surface-mechanical and oxidation behavior of electro-co-deposited Cu-Y<sub>2</sub>O<sub>3</sub> composite coating, Electrodeposition and characterization of Ni-Y<sub>2</sub>O<sub>3</sub> composite, Surf. Coat. Technol. 304 (2016) 348–358.



# **Synthesis and characterization of pulsed electrodeposited $\text{Cu-Y}_2\text{O}_3$ coating**

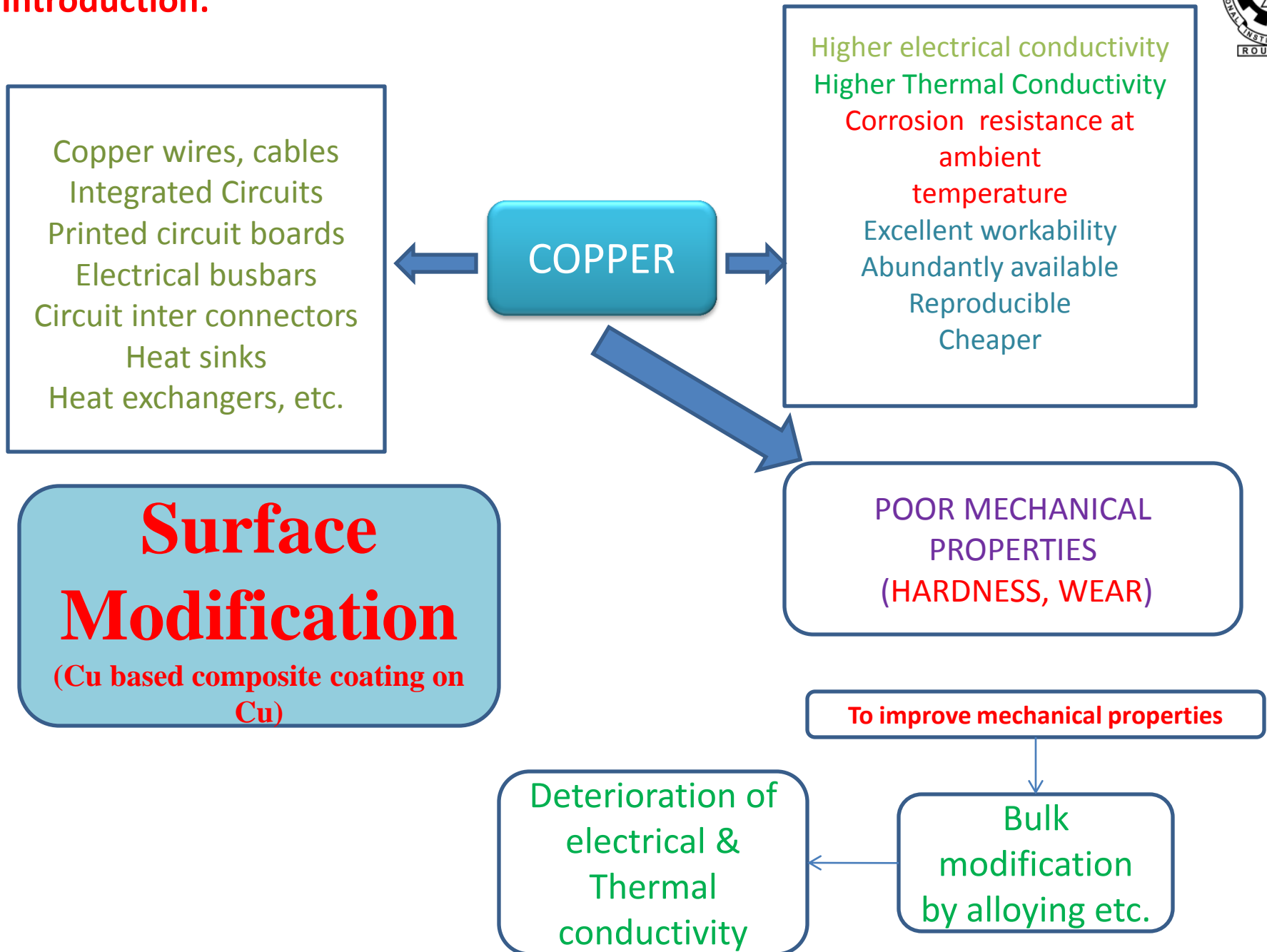
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## Introduction:



# Experimental

**Acidified Copper  
Sulfate Solution**

**Second phase  
particles  
( $Y_2O_3$ )**

**Suspension for  
Electro codeposition  
(ECD)**

**Prepared **Cu**  
Substrate**

**ECD Process (direct  
and pulsed)**

**Characterization**

**Phase  
Analysis  
(XRD)**

**Composition &  
Microstructure  
(EDS, FESEM,  
TEXTURE)**

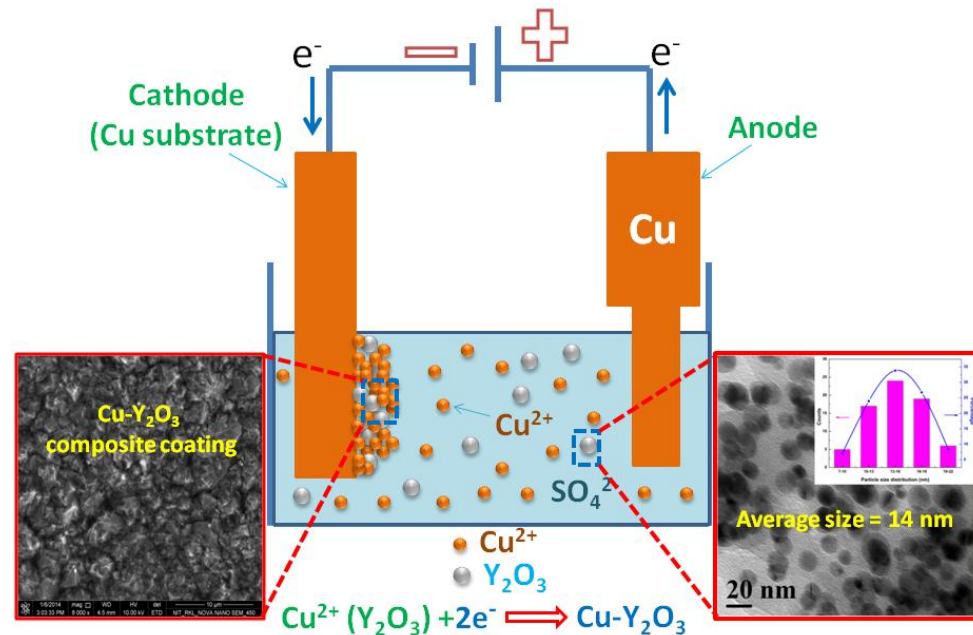
**Mechanical  
(Hardness,  
Wear)**

**Oxidation**

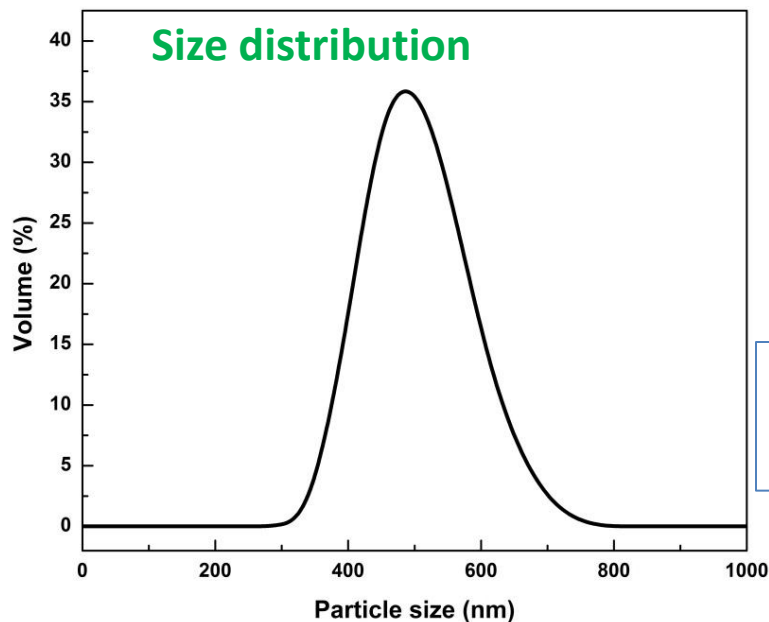
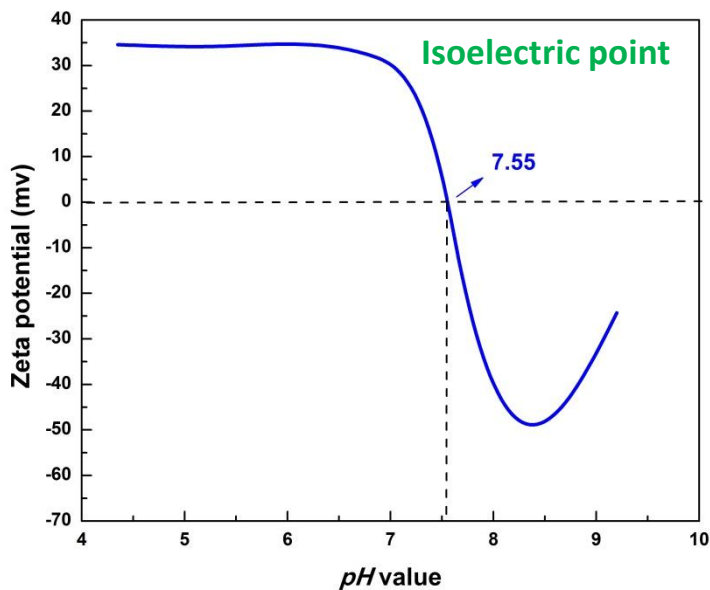
**Resistivity  
(Four probe  
Method)**

# Plating bath compositions and deposition parameters and process:

Item	Details
Electrolyte	Copper Sulfate ( $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ ): 200 gm/l, Sulfuric acid ( $\text{H}_2\text{SO}_4$ ): 50 gm/l
pH	~2.17
Current density	8 A/dm <sup>2</sup>
Temp.	Room temperature
Plating time	20 minutes
Dispersion	$\text{Y}_2\text{O}_3$ : 0, 10 and 30 gm/l
Deposition mode	DC and Pulsed DC [pulse frequency: 1, 5 and 10 kHz and duty cycle (pulse on time):30%]

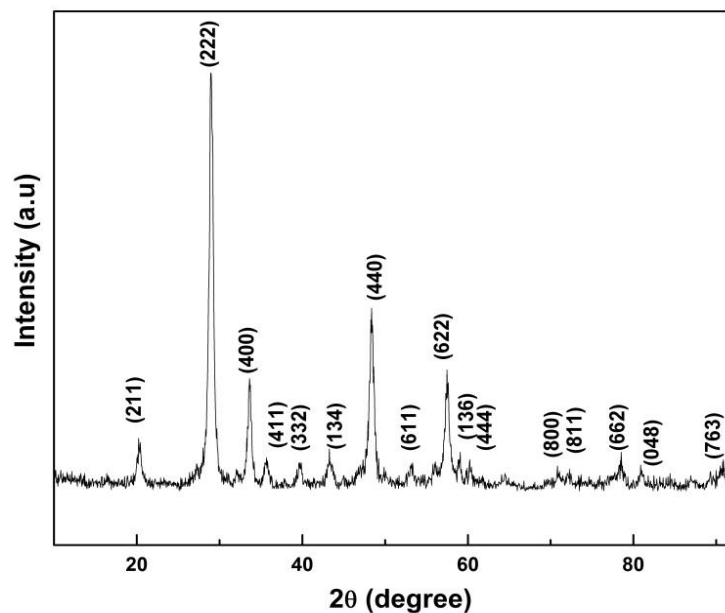
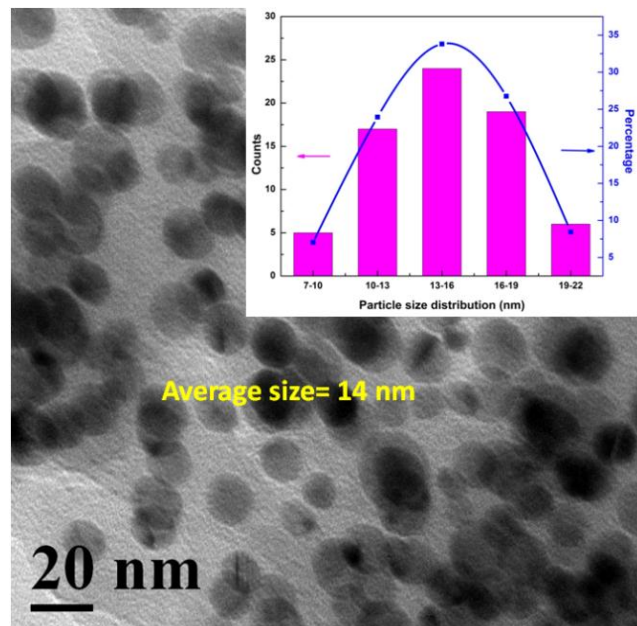


# Characterization of as procured $\text{Y}_2\text{O}_3$ powder



IEP=7.55

Avg. Size (Zeta)  
= 480nm

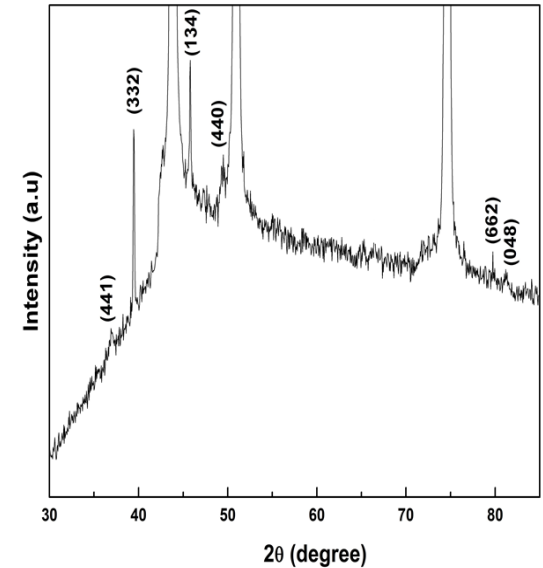
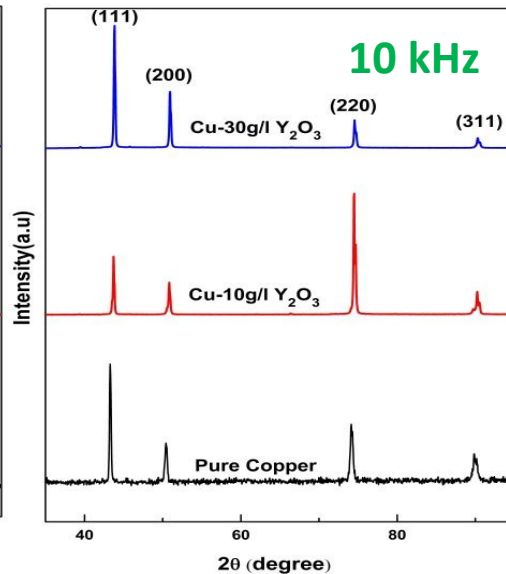
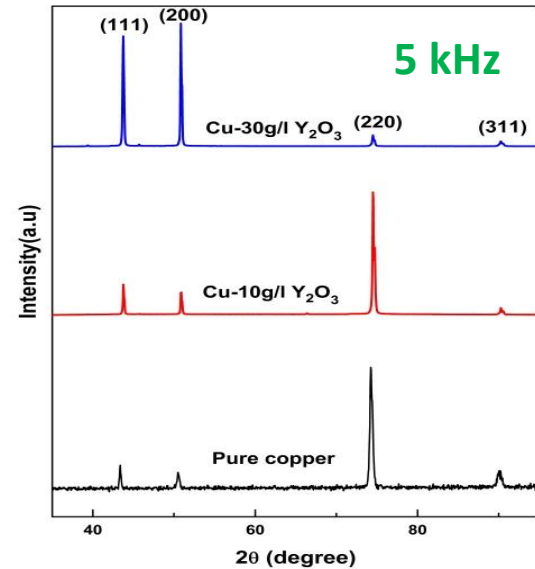
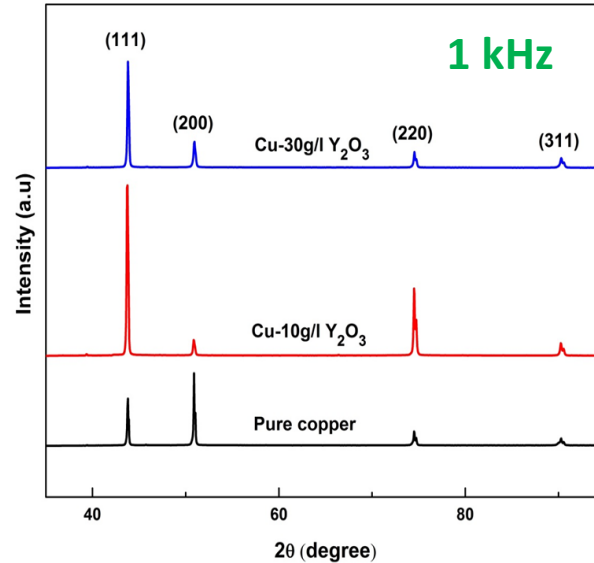
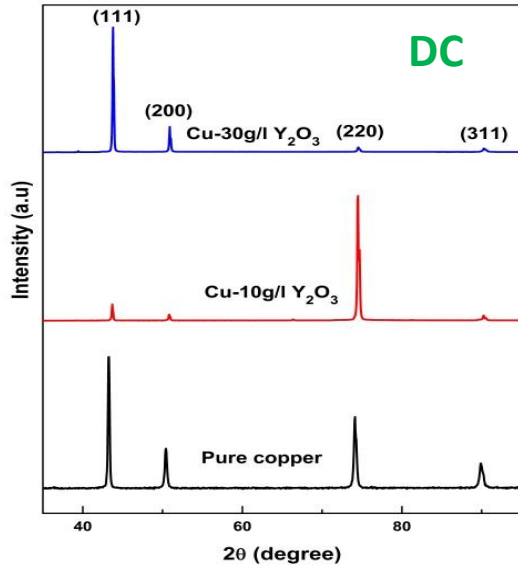


Avg. Size (TEM)  
= 14nm

BCC Crystal structure



# XRD analysis



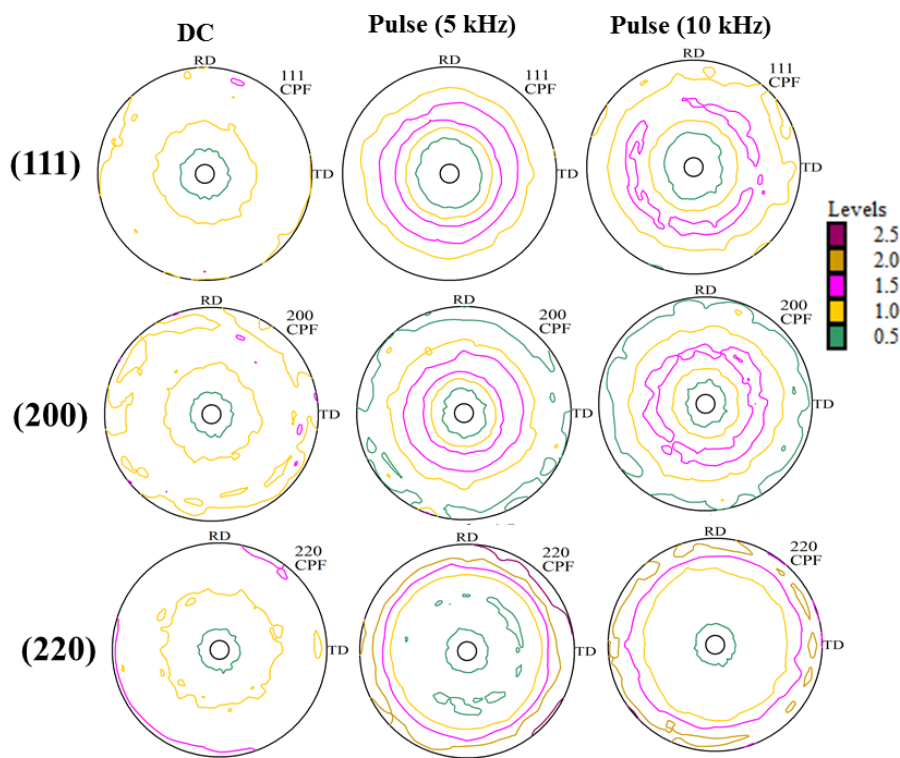
Enlarged view of XRD plot of  
Cu-30g/l  $Y_2O_3$  coating with  
10 kHz.

$Y_2O_3$  peaks are visible

Change in relative intensities of Cu peaks at  
different coating conditions

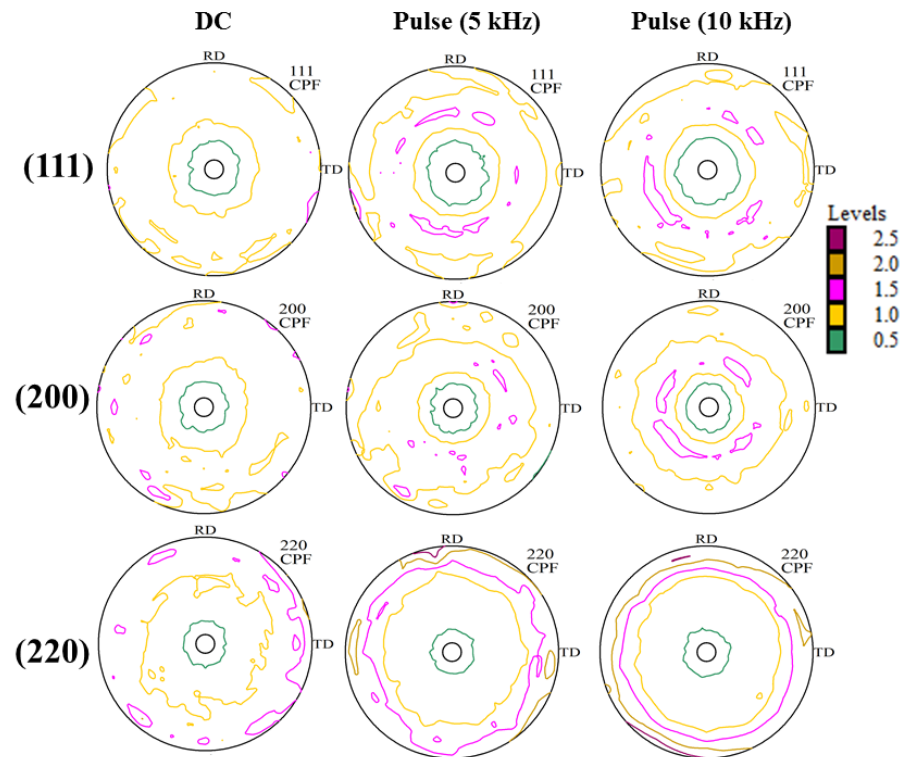
# Texture analysis

## Pure Cu



With changing deposition condition from DC to pulse frequency of 5 and 10 kHz, the initial intensity of [001], [110] and [111] orientations decreases. But the [111] orientation remained as preferred orientation

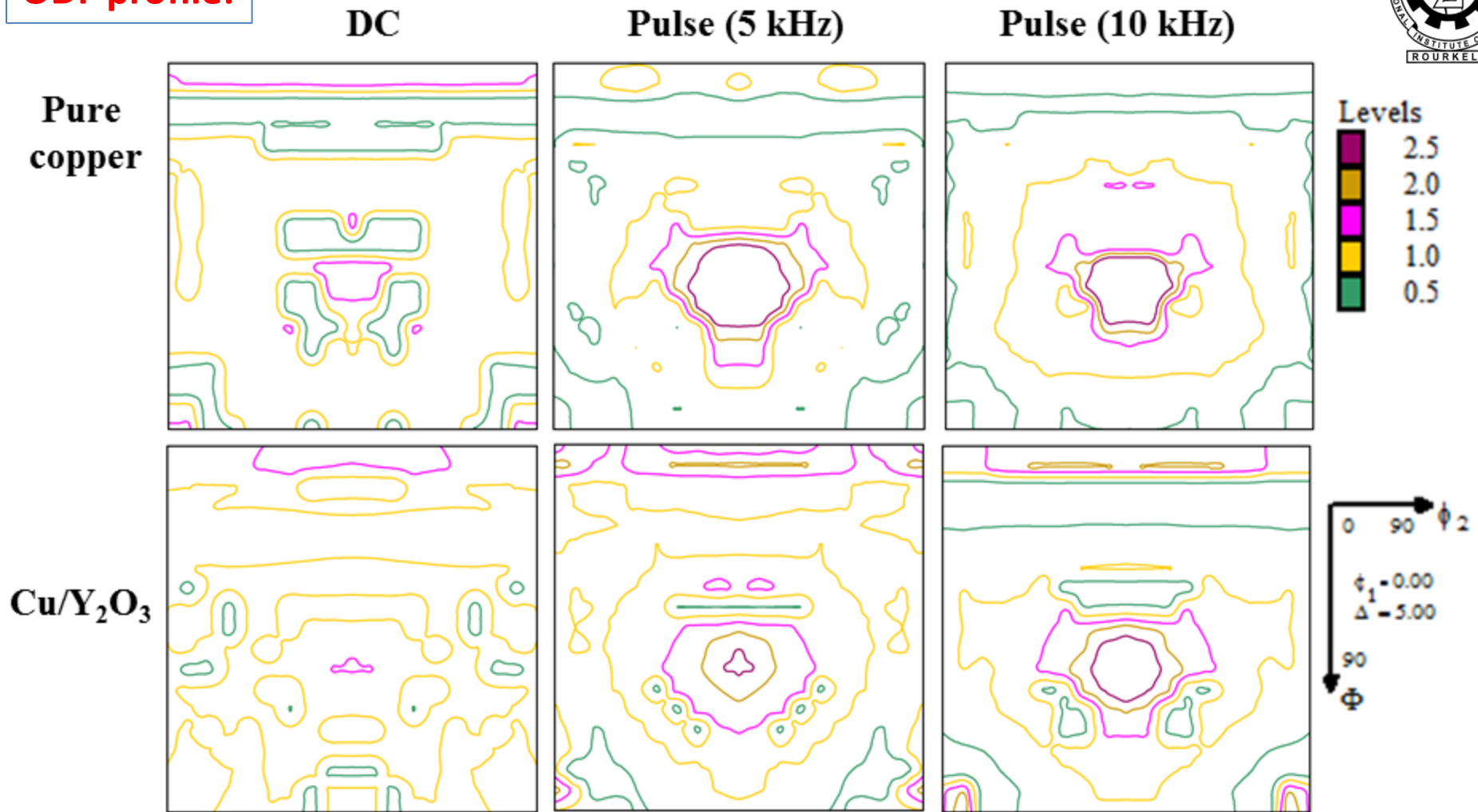
## Cu-30g/l $Y_2O_3$



Changing deposition condition from DC to 10 kHz pulse increases [111] fibre orientation and decreases other orientations ([001] and [110])



## ODF profile:

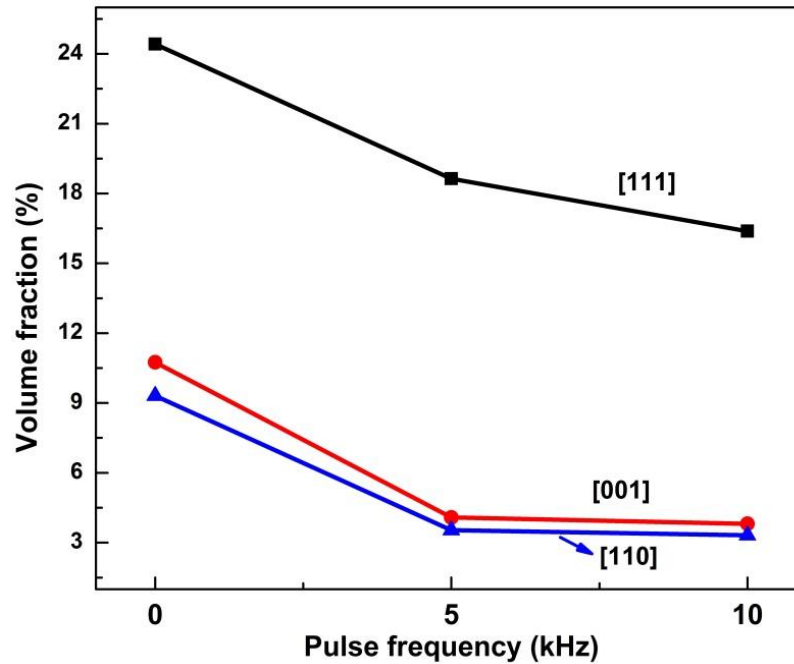


$\phi_1 = \text{constant}$  ODF figure of pure copper and Cu-30g/l Y<sub>2</sub>O<sub>3</sub> composite coatings

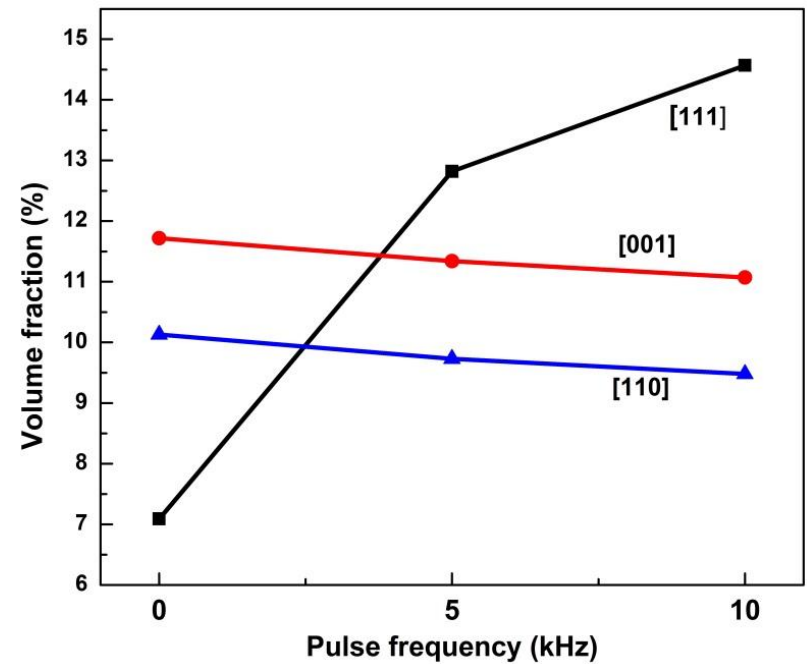
Existence of mainly [111] orientation at 10 kHz

# Orientation Volume fraction:

Pure Cu



Cu-30g/l  $Y_2O_3$



Volume fraction of different orientation confirms [111] as the main and preferred orientation.  $Y_2O_3$  addition changes the trend of orientation may be due to impact on nucleation

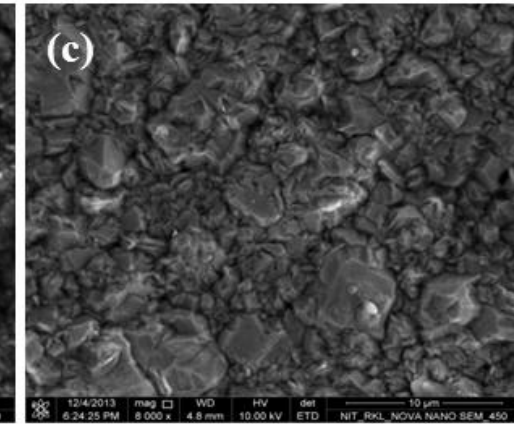
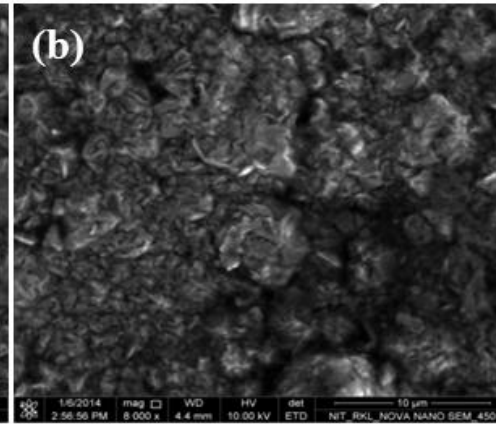
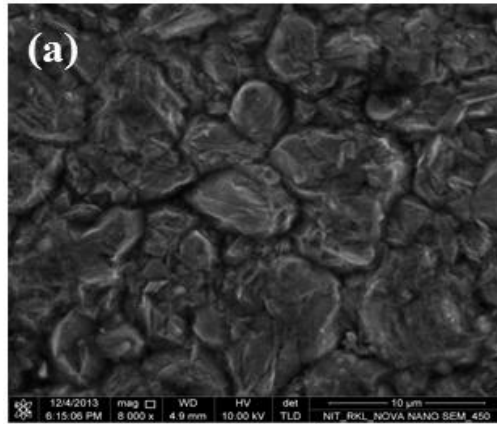
# Morphology study:

DC

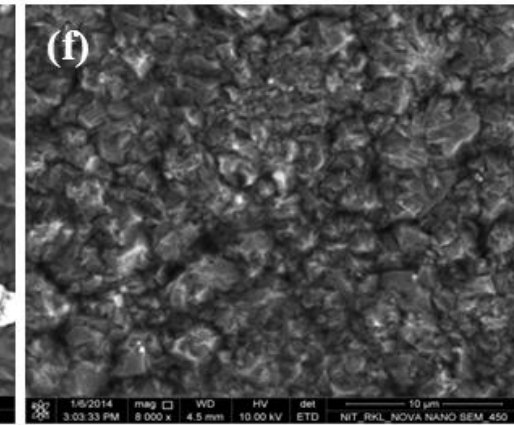
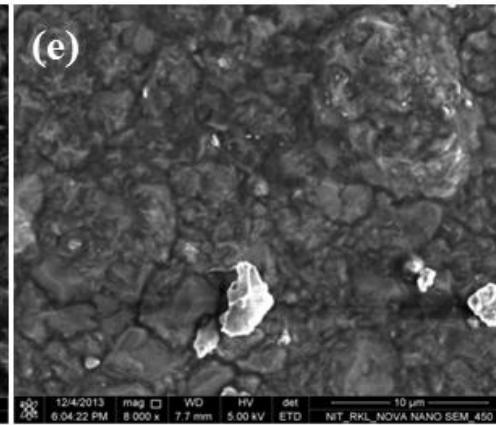
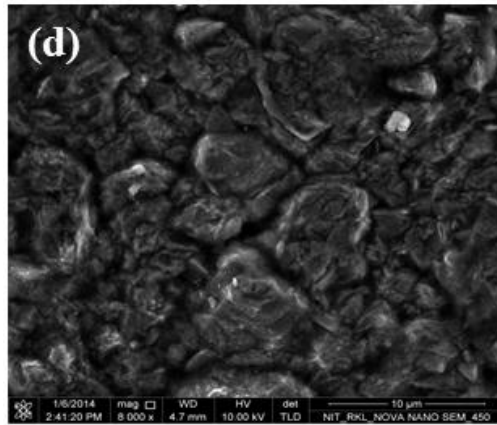
5 kHz

10 kHz

Cu-10g/l  $Y_2O_3$



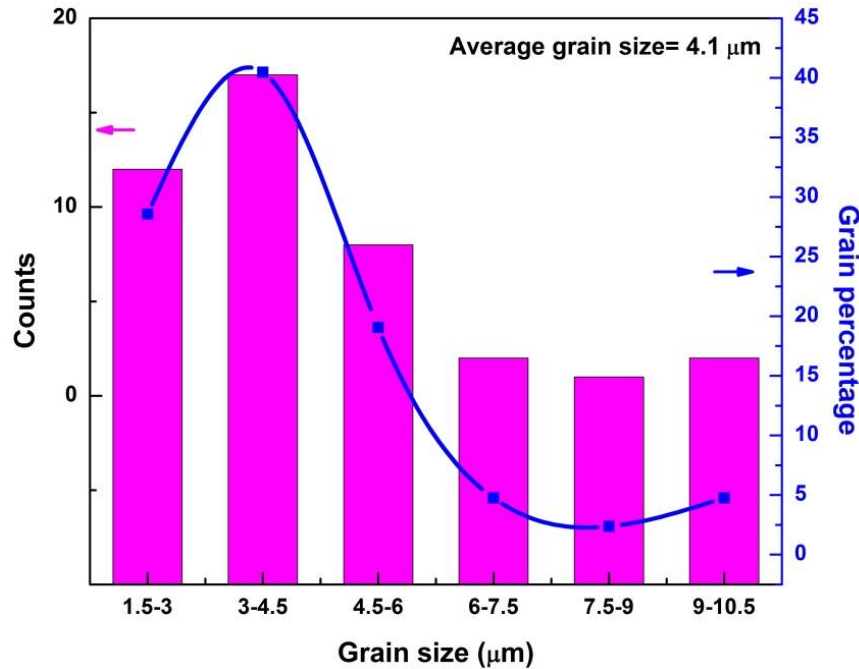
Cu-30g/l  $Y_2O_3$



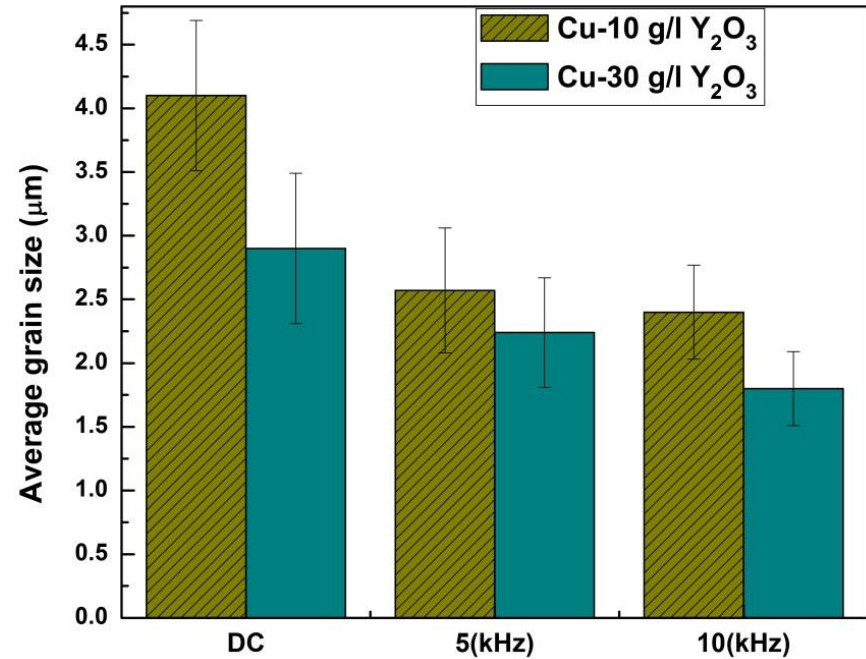
- DC deposition shows large granular structure with well-defined grain boundaries
- The coating deposited with highest pulse frequency (10kHz) exhibited well-refined granular structure with more uniform and finer matrix.

# Size distribution and average grain size

Cu-10g/l  $Y_2O_3$  ( DC )



Average grain size



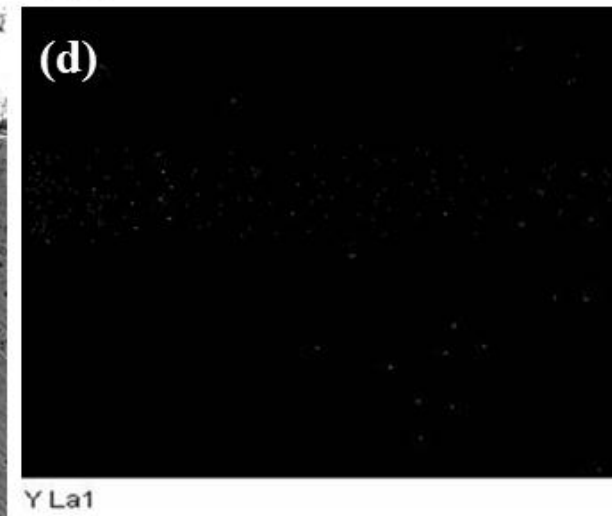
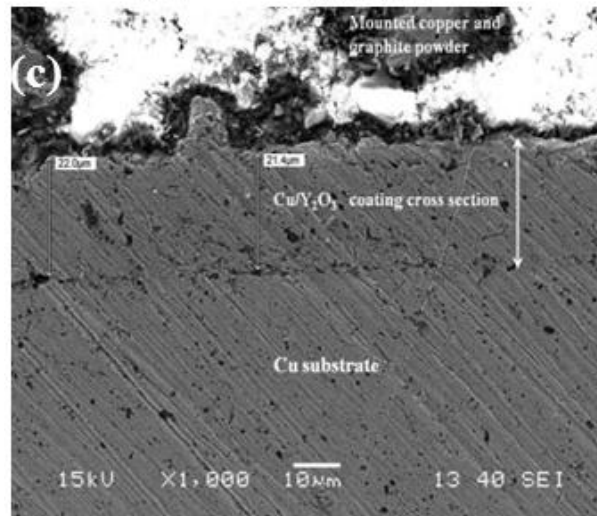
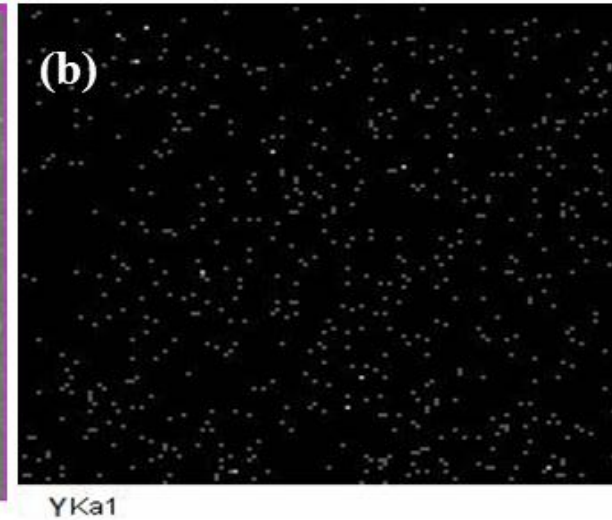
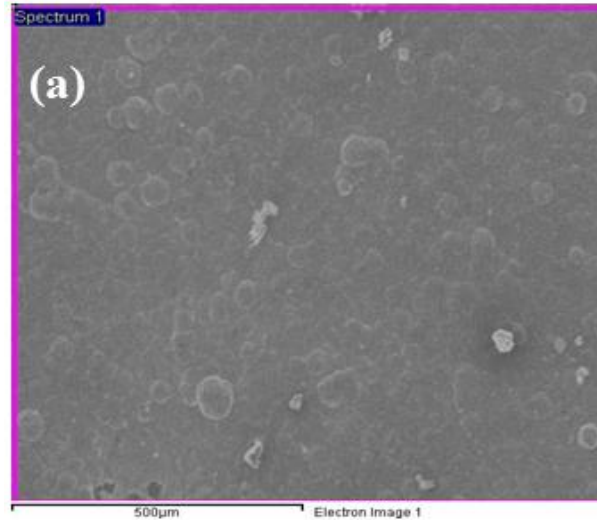
Decrease in average grain size with increase in second phase particle dispersion and pulse frequency

Published in: *Surface and Coating Technologies*, 304 (2016) 348–358; Basu et. al

# Elemental mapping

Cu-30g/l  $Y_2O_3$  (10 kHz)

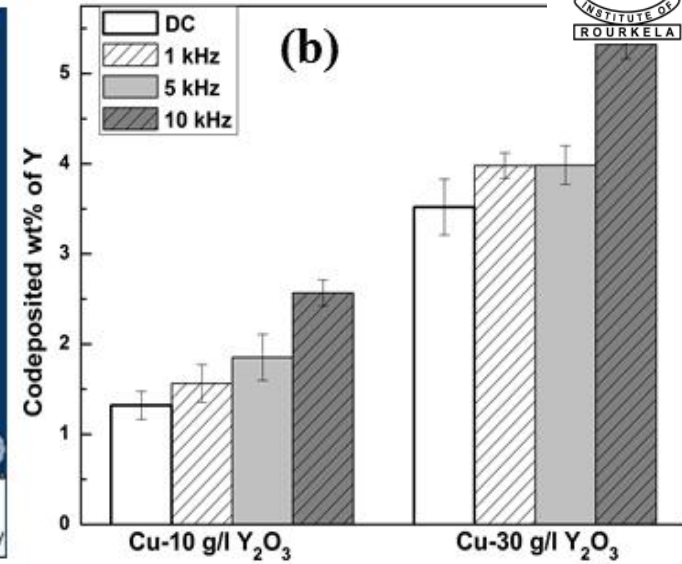
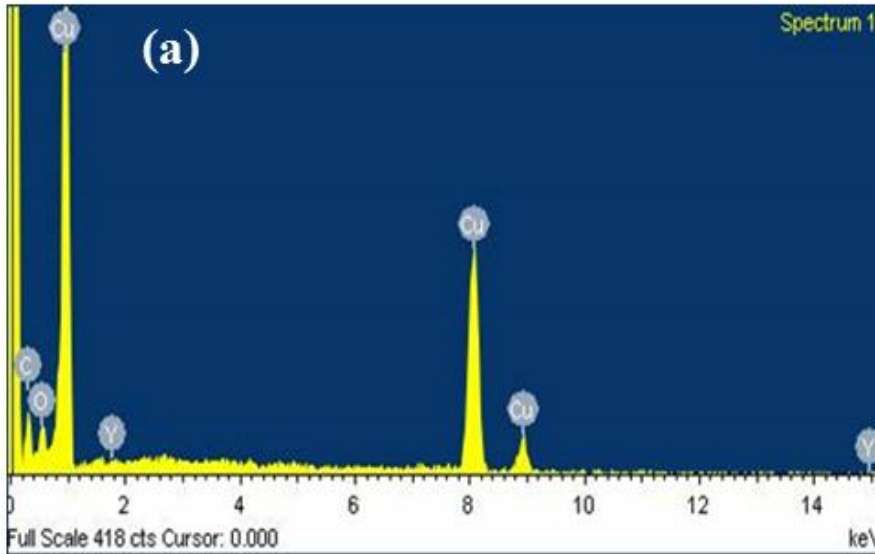
Uniform deposition of  $Y_2O_3$  throughout the coating and cross-section





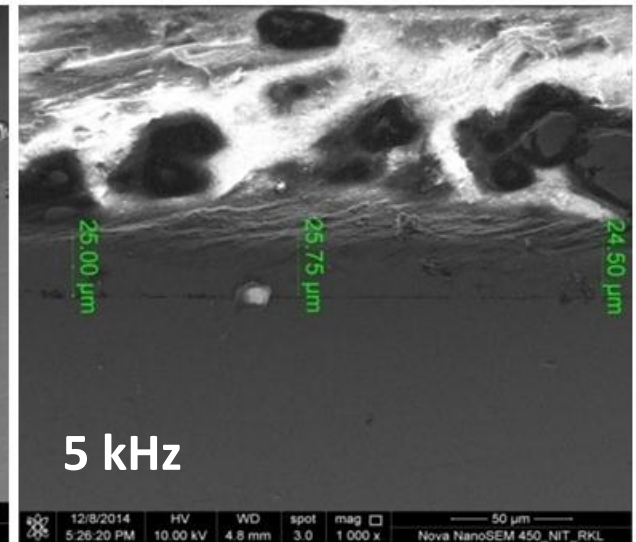
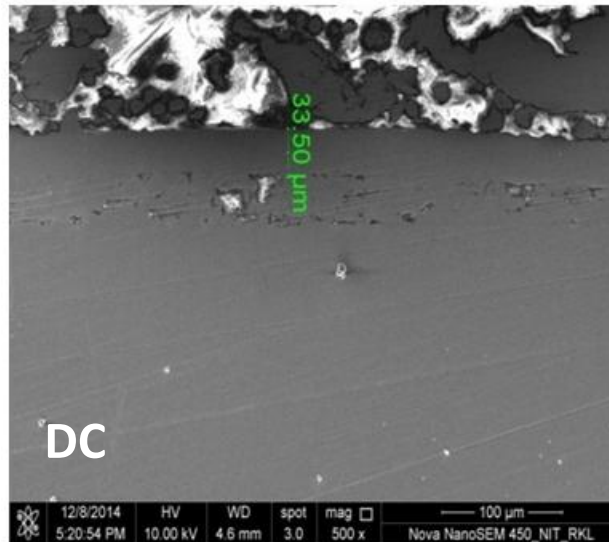
# EDS spectra

Cu-30g/l  $Y_2O_3$ (10 kHz)



- Wt.% of  $Y_2O_3$  in the coating increases with pulse frequency and  $Y_2O_3$  dispersion.
- Thickness decreases with increasing pulse frequency.

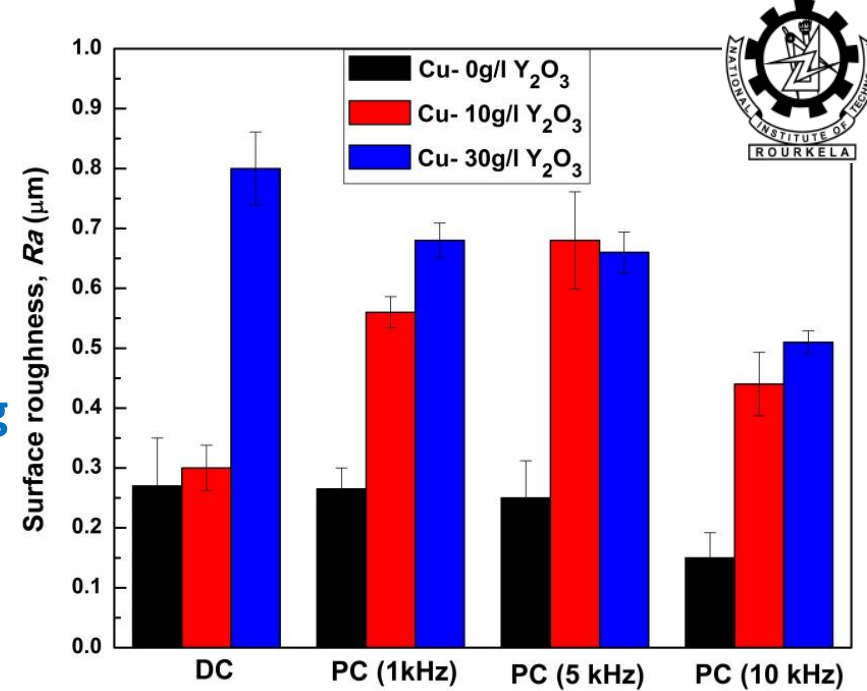
Cu-30g/l  $Y_2O_3$





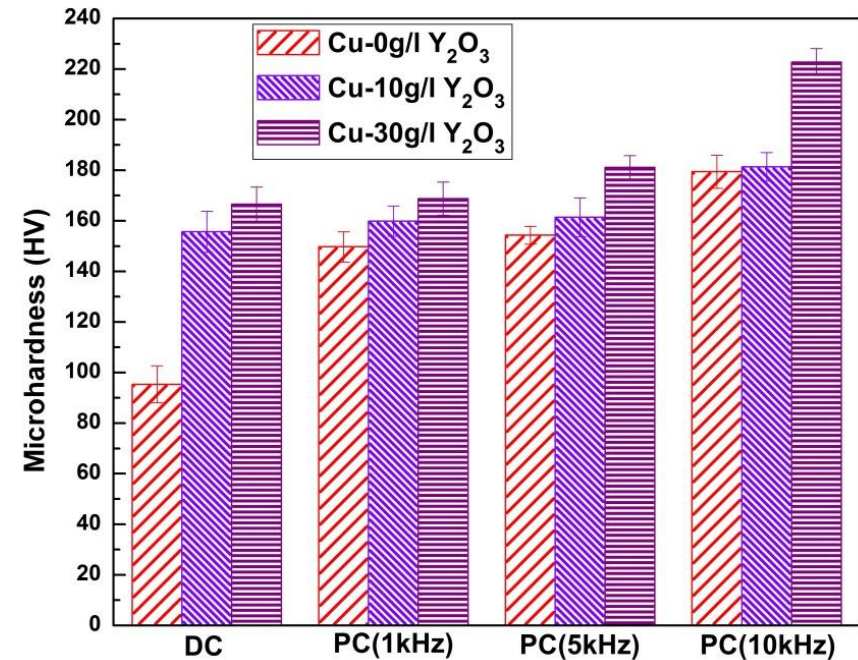
# Surface Roughness

- $R_a$  0.14 to 0.8  $\mu\text{m}$
- Surface roughness increases with increasing  $\text{Y}_2\text{O}_3$  content except 5 kHz



# Microhardness

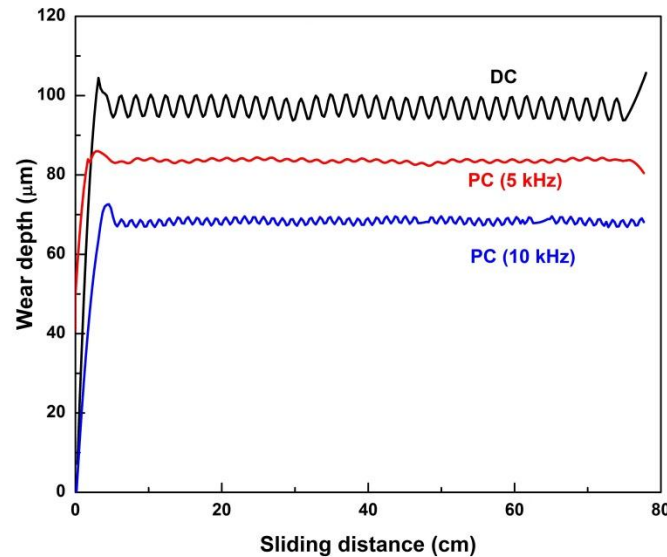
Hardness increased with increasing  $\text{Y}_2\text{O}_3$  content as well as pulse frequency.



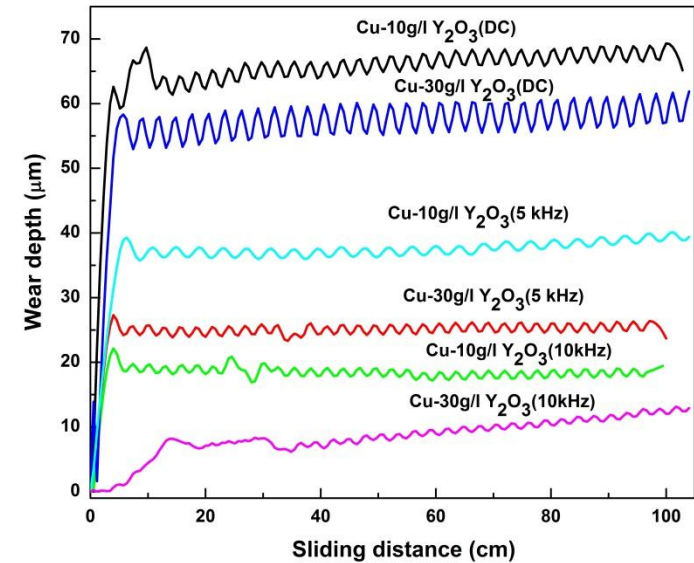
# Wear study

- Wear resistance improvement by pulsing and  $Y_2O_3$  content

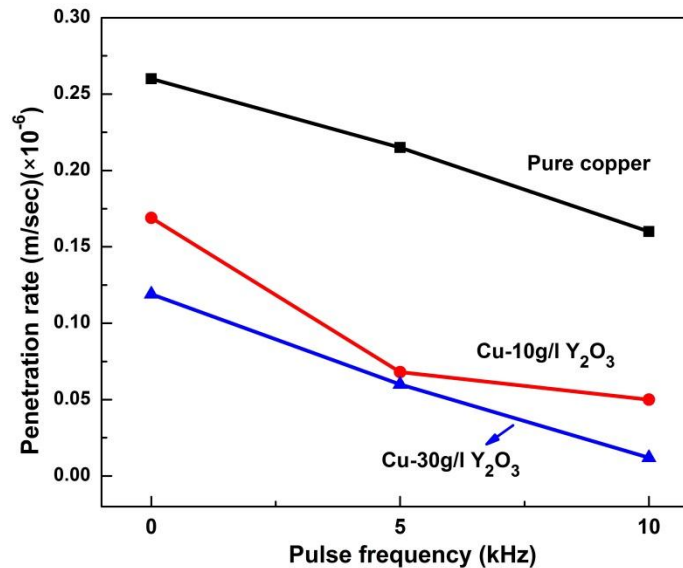
## Pure Cu



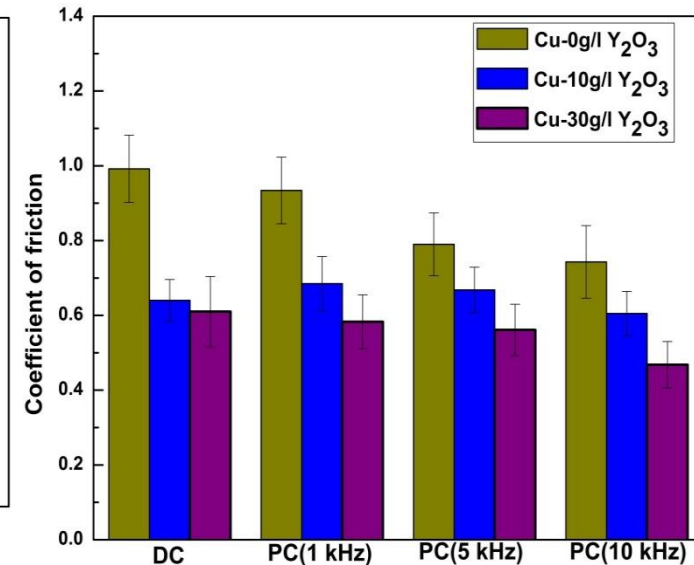
## Cu- $Y_2O_3$



## Wear rate



## Coefficient friction



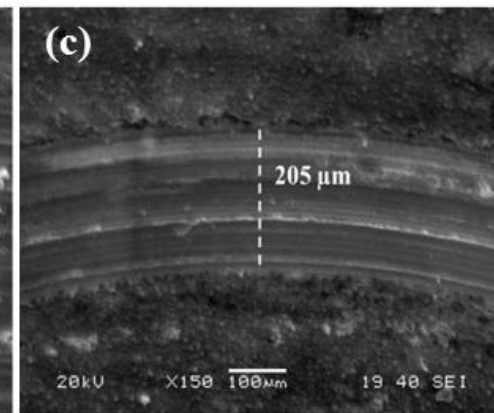
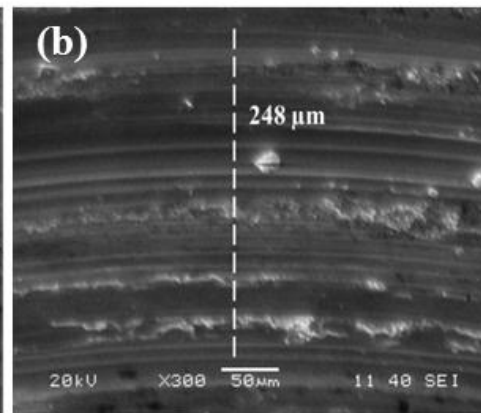
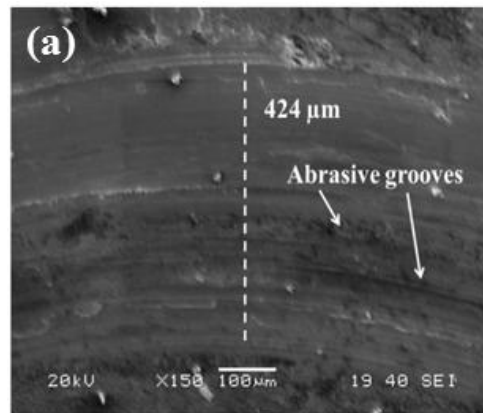
- Friction coefficient is decreases with  $Y_2O_3$  content and pulse frequency

## Wear track

Pure Cu (10 kHz)

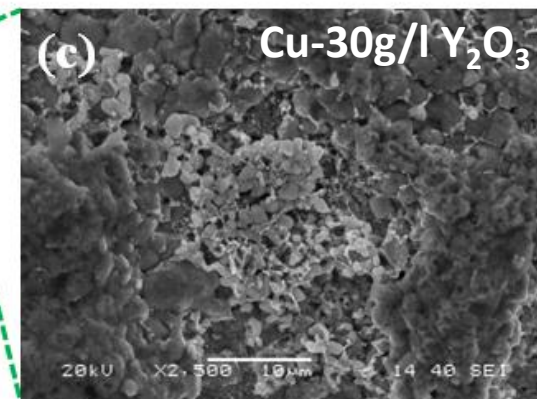
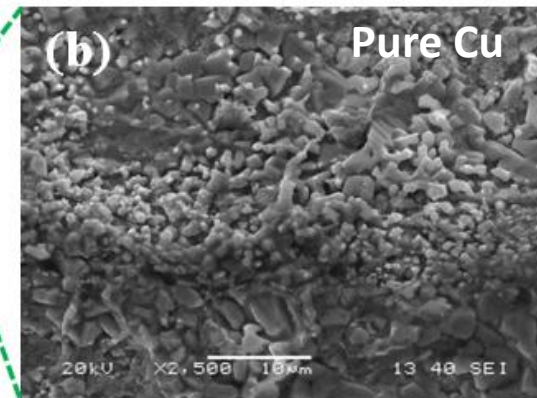
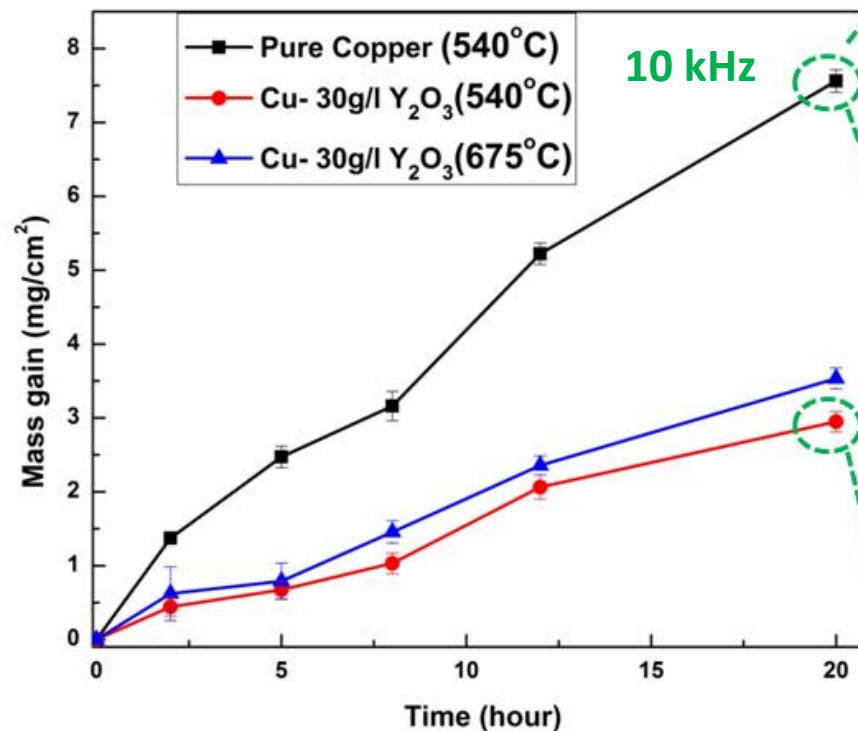
Cu-10g/l  $Y_2O_3$  (10 kHz)

Cu-30g/l  $Y_2O_3$  (10 kHz)



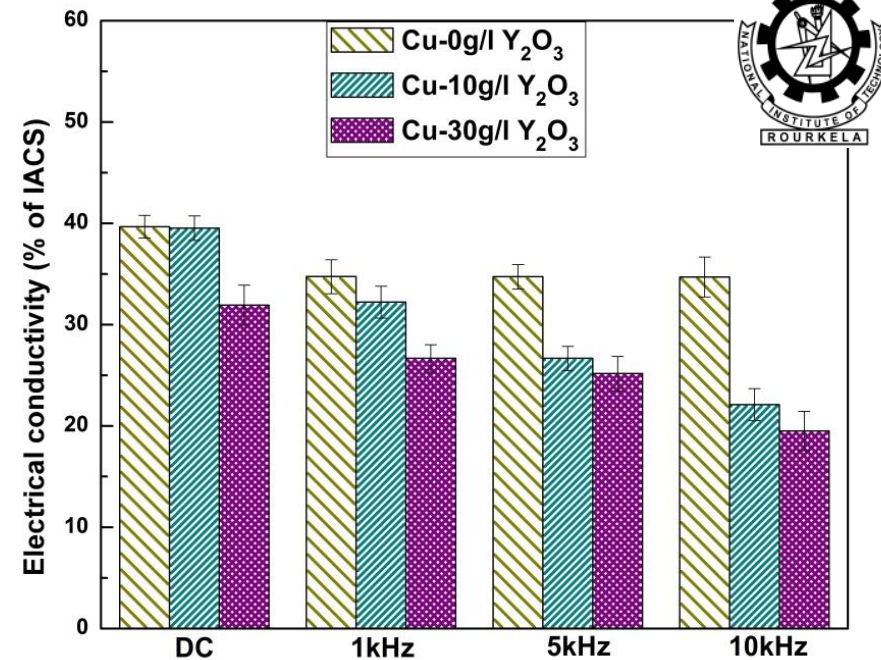
## Oxidation study

- Composite coating exhibited better oxidation resistance



## Electrical conductivity

- Decrease in conductivity with increase in pulse frequency in spite of (111) orientation development; can be attributed to dispersed  $Y_2O_3$  particle, finer grain size.



### After oxidation

Pure Cu- (13-24 %)



Cu- $Y_2O_3$ - (1-7.7 %)





## Conclusions

- Texture study confirms [111] as the main fibre orientation.
- Higher hardness and better wear resistance properties of codeposited coatings were attributed to finer matrix and more amount of embedded  $Y_2O_3$  particles resulted by pulsing mode with higher pulse frequency conditions as well as some texture effect.
- Better oxidation resistance was found in composite coatings due to better microstructure.
- Minor decrease in electrical conductivity of composite coatings was observed due to presence of  $Y_2O_3$  and finer matrix.

# Acknowledgement

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# Thank You!