

Raman spectroscopy as an analytical tool for characterization of fluoroquinolones

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ABSTRACT

The aim of this paper is to produce the basis for the Raman spectroscopy as a promising new official method for identification of fluoroquinolone antibacterials. Raman spectroscopic analysis data for three pure fluoroquinolone antibacterials like Norfloxacin, Ciprofloxacin and Ofloxacin, have been performed by a low resolution portable Raman Spectrometer (Raman system R-3000) using 785 nm solid state diode laser and having fibre optic sampling probe. The Raman spectra are collected over the wave number range from 140 to 2500 cm^{-1} at room temperature. From our overall Raman spectroscopic analysis, it has been found that those fluoroquinolones show prominent characteristic peaks for pyridone nucleus, piperazinyl ring, carboxylic acid and fluorine substituent. In addition, the characteristic peaks at 872.7, 771.47 and 797.5 cm^{-1} for Norfloxacin, Ciprofloxacin and Ofloxacin, respectively, are assigned to fully symmetric vibrational normal mode of the molecule containing the C-F bond. Since the important characteristic groups are available in all the cases, their Raman spectral pattern is more or less similar. From the above-mentioned information, it can be concluded that Raman spectroscopic analysis, a non-destructive method with minimal water interference, may also be used as a useful tool for the identification of compounds, specially the fluoroquinolones.

Key words: Norfloxacin, Ciprofloxacin, Ofloxacin, Raman spectroscopy

INTRODUCTION

Fluoroquinolones are widely used synthetic broad-spectrum antibacterial agents. They form a group of antimicrobial agents with different chemical structures and spectra of activity. The general molecular structure of fluoroquinolones like Norfloxacin, Ciprofloxacin and Ofloxacin consists of 1-substituted-1,4-dihydro-4-oxopyridine-3-carboxylic moiety combined with an aromatic or heteroaromatic ring (Fig. 1). The benzopyridone nucleus (quinolone) proved to be more responsive to chemical manipulation in order to enhance antibacterial potency. The subsequent discovery of fluorine atom and piperazinyl ring on the quinolone nucleus revolutionized the chemistry and clinical importance of fluoroquinolones [1,2].

While Raman spectroscopy has long been recognized as a valuable research technique, only very recently this technique has emerged as an important analytical tool across a number of industries and applications. Due to its sensitivity, high information content, and non-destructive nature, Raman spectroscopy is now used in many applications across the fields of chemistry, biology, geology, pharmacology, forensics, pharmaceuticals, materials science, and failure analysis. Moreover, FTIR and Raman spectroscopy are used as complementary techniques, because each method looks at different aspects of a given sample [3]. Since dispersive Raman spectroscopy operates at visible wavelengths, its signal is much stronger. In addition, the technique is more sensitive and it delivers higher spatial resolution. This makes dispersive Raman as the best choice for small particles and minor component analysis [4].

It has also been found that Raman spectroscopy can identify fluoroorganic compounds in liquid, gaseous, crystalline or amorphous solid states. The method is more sensitive than other methods for detecting the C-F bond of fluoroorganic compounds, permitting detection of fluoroorganic compounds at 10^{-3} - 10^{-6} g/L concentrations level. Raman emission for C-F bond is centered on 500-800 cm^{-1} , which is characteristic for fluoroorganic compounds. On the other hand, the frequency range for C-F bond, FTIR absorption is from 1000 cm^{-1} to 1400 cm^{-1} . But these FTIR absorptions are subject to interference by other functional groups present in respect to fluoroorganic compounds. On the other hand, in case of Raman spectroscopy, the C-F bond vibration assigned as follows: 710-785 cm^{-1} for trifluoromethyl groups; 726 cm^{-1} for 1-bromoperfluorooctane; 730 cm^{-1} for perfluorodecanoic acid; 750 cm^{-1} for triperfluoropropylamine; and 730 cm^{-1} for 1,3,5-tris-(trifluoromethyl)benzene. Most of these compounds have an aromatic ring, so resonance Raman can be observed [5]. As fluoroquinolones are aromatic fluoroorganic compounds having C-F groups as a part of molecular structure, this type of C-F bond can also be identified by Raman spectroscopy.

Till now, Norfloxacin, Ciprofloxacin and Ofloxacin, and their formulations are included in USP XXIV [6]. Moreover, Norfloxacin and Ciprofloxacin, and their formulations are included in BP [7] and IP [8]. However, the Raman spectroscopic analysis method has not been suggested in the USP, BP and IP for identification of fluoroquinolones, but infrared absorption, thin layer chromatography, high performance liquid chromatography, polarographic methods, etc. have been recommended. A review of the literature reveals that very little attention has been paid to the Raman spectroscopic analysis for the identification and determination of fluoroquinolone antibacterials. While FTIR is sensitive to functional groups and highly polar bonds, Raman spectroscopy is more sensitive to backbone structures and symmetric bonds.[9] Taking into consideration of above positive aspects of Raman spectroscopic analysis, we have used this as an analytical tool for the identification of fluoroquinolones like Norfloxacin, Ciprofloxacin and Ofloxacin.

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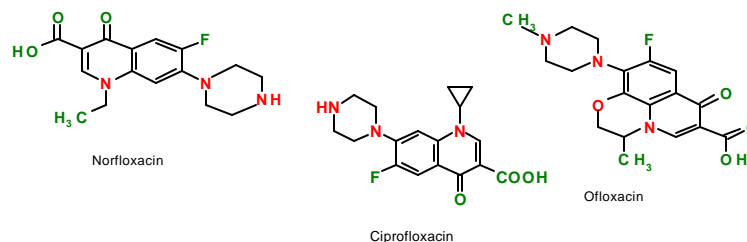


Figure 1. Molecular Structures of some Fluoroquinolones

MATERIALS AND METHODS

Materials:

Pure samples of Norfloxacin (C16H18FN3O3), Ciprofloxacin (C17H18FN3O3) and Ofloxacin (C18H20FN3O4) were provided as gift samples by Dr. Reddy's Research Foundation, Hyderabad, India. The drugs were of 99.8% to 98.0% purity.

Methods:

The Raman system R-3000 instrument, manufactured by Raman systems INC.USA, is a low resolution portable Raman Spectrometer using a 785 nm solid state diode laser adjusted to deliver 250 mw to the sample having spectral resolution 10 cm^{-1} and 12 v dc/5A power supplies and USB connectivity. This Raman Spectrometer is having fibre optic sampling probe with safety shutter and automatic focusing caps for both solid and liquid samples. The solid powder samples i.e., pure drugs were enclosed in plastic poly bags and tested directly at room temperature. The interference of the outside light was also prohibited to prevent photon shot noise. The spectra were collected over the wave number range from 140 to 2500 cm^{-1} .

RESULTS

By Raman spectroscopy of Norfloxacin, the prominent Raman shifts have been observed at 485.6, 872.7, 1418.5 and 1655.1 cm^{-1} (Fig. 2). The Raman shifts at 485.6 cm^{-1} indicates strong bending vibration of C-C of the aliphatic chain and C-N stretching vibration of piperazinyl group [10,11,12]. The band at 872.7 cm^{-1} represents the symmetric stretching vibration of C-F group [13]. The peak at 1418.5 cm^{-1} is due to symmetric stretching vibration of O-C-O group of carboxylic acid and methylene deformation mode of the piperazinyl group [14]. A band at 1655.1 cm^{-1} is due to symmetric stretching of the carbonyl group $\nu_{\text{C=O}}$ of the pyridone moiety, the stretching vibration of (C-C) aromatic ring chain. In addition, it (peak at 1655.1 cm^{-1}), also indicates the N-H₂ scissoring of piperazinyl group [10,14,15,16,17] (Table 1a).

In case of Ciprofloxacin, the prominent Raman shifts have been observed at 484.22, 771.47, 1411.63 and 1655.11 cm^{-1} (Fig. 3). The Raman shifts at 484.22 cm^{-1} indicates strong bending vibration of C-C of the aliphatic chain of cyclopropyl group and C-N stretching vibration of piperazinyl group [10,11,12]. The band at 771.47 cm^{-1} represents the symmetric stretching vibration of C-F group [5]. The peak at 1411.63 cm^{-1} is due to symmetric stretching vibration of O-C-O group of carboxylic acid and methylene deformation mode of the piperazinyl group [14]. A band at 1655.11 cm^{-1} is due to symmetric stretching of the carbonyl group $\nu_{\text{C=O}}$ of the pyridone moiety, the stretching vibration of (C-C) aromatic ring chain [18]. In addition, it (peak at 1655.11 cm^{-1}), also indicates the N-H₂ scissoring of piperazinyl group [10,14,15,16,17,18] (Table 1b).

By Raman spectroscopy of Ofloxacin, the prominent Raman shifts have been observed at 518.4, 797.5, 1419.8 and 1649.6 cm^{-1} (Fig. 4). The Raman shift at 518.4 cm^{-1} represents the bending vibration of aliphatic carbon atom, C-N stretching vibration of piperazinyl group and O-H torsional vibration of carboxylic acid [10,11,12,14]. The band at 797.5 cm^{-1} represents the symmetric stretching vibration of C-F group [5]. The peak at 1419.8 cm^{-1} is due to symmetric stretching vibration of O-C-O group of carboxylic acid and methylene deformation mode of the piperazinyl group [14]. A band at 1649.6 cm^{-1} is due to symmetric stretching of the carbonyl group $\nu_{\text{C=O}}$ of the pyridone moiety, the stretching vibration of (C-C) aromatic ring chain. In addition, it (peak at 1649.6 cm^{-1}), also indicates the N^+H_2 scissoring of piperazinyl group [10,14,15,16,17] (Table 1c).

Table 1. Prominent Raman Shifts of Some Fluoroquinolones

a) Prominent Raman Shifts of Norfloxacin	
Raman Shifts(cm^{-1})	Functional Groups / Vibrations
485.6	Strong $\delta(\text{CC})$ aliphatic chain and C-N stretching vibration
872.7	Symmetric vibration of C-F bond
1418.5	$\nu_{\text{O-C-O}}$ and methylene deformation of the piperazinyl group
1655.1	ν_{s} of C=O group of pyridone moiety and N^+H_2 scissoring of piperazinyl group
b) Prominent Raman Shifts of Ciprofloxacin	
Raman Shifts(cm^{-1})	Functional Groups / Vibrations
484.22	Strong $\delta(\text{CC})$ aliphatic chain and C-N stretching vibration
771.47	Symmetric vibration of C-F bond
1411.63	ν_{s} O-C-O and methylene deformation of the piperazinyl group
1655.11	ν_{s} of C=O group of pyridone moiety and N^+H_2 scissoring of piperazinyl group
c) Prominent Raman Shifts of Ofloxacin	
Raman Shifts(cm^{-1})	Functional Groups / Vibrations
518.4	Strong $\delta(\text{CC})$ aliphatic chain, C-N stretching vibration and O-H torsional vibration
797.5	Symmetric vibration of C-F bond
1419.8	ν_{s} O-C-O and methylene deformation of the piperazinyl group
1649.6	ν_{s} of C=O group of pyridone moiety and N^+H_2 scissoring of piperazinyl group

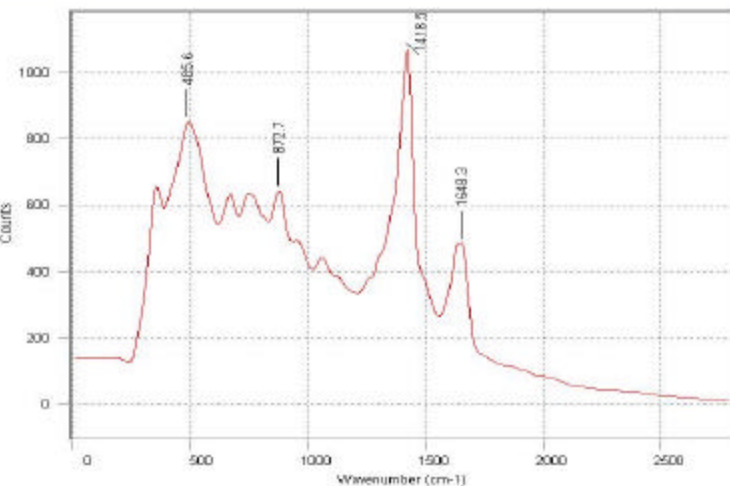


Figure 2: Raman Shifts of Norfloxacin

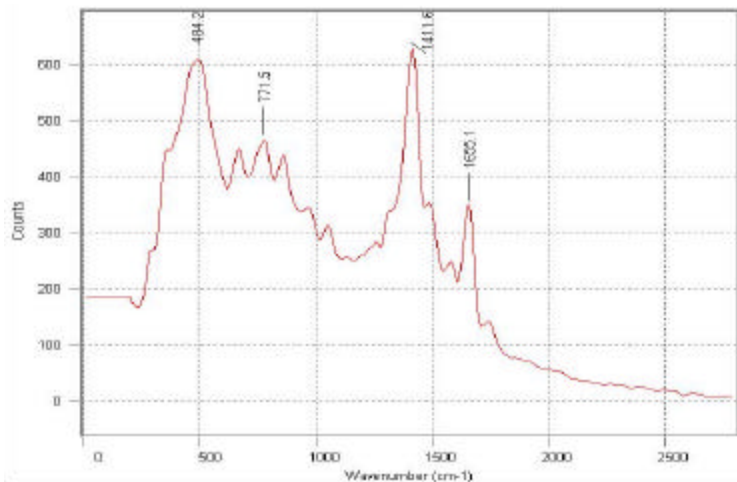


Figure 3: Raman Shifts of Ciprofloxacin

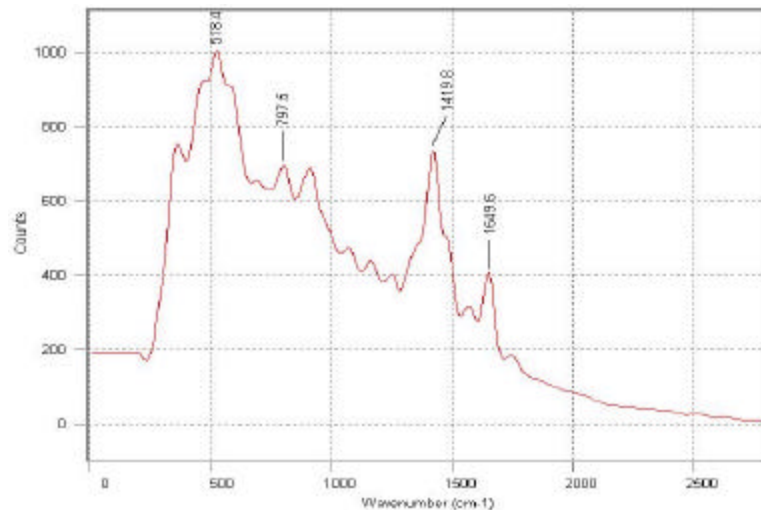


Figure 4: Raman Shifts of Ofloxacin

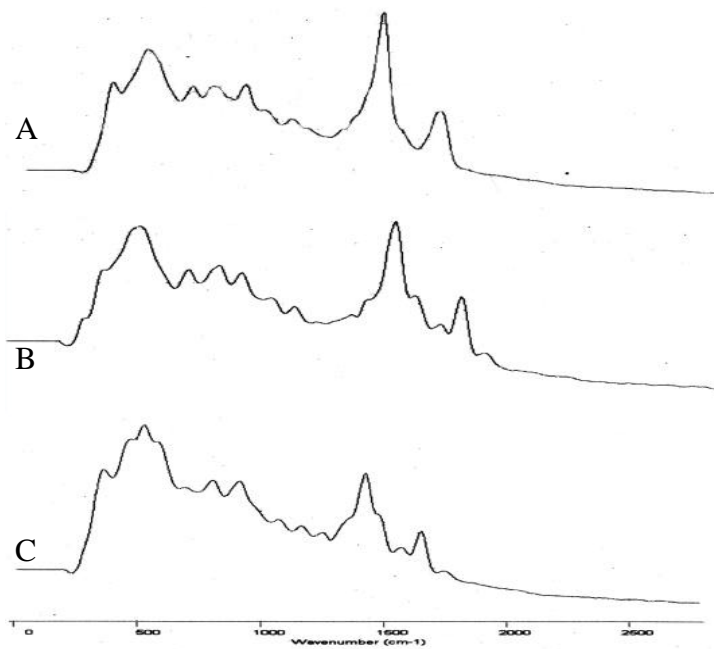


Figure 5: Comparative Raman Shifts of Fluoroquinolones

For our convenience, the comparative Raman spectra have also been represented in Fig. 5.

DISCUSSION

It is known that the backbone of fluoroquinolone antibacterials consists of a 1-substituted-1,4-dihydro-4-oxopyridine-3-carboxylic moiety combined with an aromatic or heteroaromatic ring. In addition, as has been mentioned earlier, the quinolone nucleus as well as fluorine atom and piperazinyl ring on the quinolone nucleus are proved to be more responsive to their antibacterial potency [1]. From our overall Raman spectroscopic analysis, it has been found that Norfloxacin, Ciprofloxacin and Ofloxacin also show prominent characteristic peaks for pyridone nucleus, piperazinyl ring and fluorine substituent. Moreover, the presence of carboxylic acid moiety indicates the symmetric stretching vibration of O-C-O group. The piperazinyl group characterized by methylene deformation and N^+H_2 scissoring due to resonating structure of the aromatic ring has commonly been detected in our study [14] (Table 1).

Several scientists have suggested that the compounds having C-F group along with resonating aromatic ring structure can be characterized by Raman spectroscopy, when Raman spectra are within 500-800 cm^{-1} [5]. From our investigation, the characteristic peaks at 872.7, 771.47 and 797.5 cm^{-1} for Norfloxacin, Ciprofloxacin and Ofloxacin, respectively, are assigned to fully symmetric vibrational normal mode of the molecule containing the C-F bond. Since the important characteristic groups are available in cases of all the drugs, their Raman spectral pattern is more or less similar (Fig. 5).

Considering the above-mentioned information, it can be concluded that Raman spectroscopic analysis, a non-destructive method with minimal water interference, may also be used as a useful tool for the identification of compounds, specially the fluoroquinolones. So, this study

may serve as the basis for the Raman spectroscopy to be considered as a promising new official method for identification of fluoroquinolone antibacterials.

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